

### Mole Conversions Worksheet

1. How many moles of magnesium is  $3.01 \times 10^{22}$  atoms of magnesium?

0.05 moles

2. How many molecules are there in 4.00 moles of glucose,  $C_6H_{12}O_6$ ?

$2.408 \times 10^{24}$

3. How many moles are  $1.20 \times 10^{25}$  atoms of phosphorous?

19.93 moles

4. How many atoms are in 0.750 moles of zinc?

$4.515 \times 10^{23}$

5. How many molecules are in 0.400 moles of dinitrogen pentoxide?

$2.408 \times 10^{23}$

11. Find the mass in grams of  $2.00 \times 10^{23}$  molecules of  $C_{15}H_{13}O$  (called penguinone because its structure looks like a penguin).

50.17 g

12. Determine the mass, in grams, of 2.6 moles of angelic acid,  $C_5H_8O_2$ .

260 g

# Mass Mole Conversion Handout Answer Key

**Scott C. Dulebohn**



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