
AQA a level physics paper 1 100% SOLUTIONS REVISED 2023//2024

What is meant by work function (2) - ANSWER Minimum energy required by an electron to escape from (1) a metal surface (1)

What is meant by ionisation energy? (2) - ANSWER Minimum energy required to remove an electron from an atom (1) from the ground state (1)

Why do emitted electrons have a kinetic energy that varies up to a maximum value? (4)
- ANSWER Energy of a photon is fixed/constant (1)
Energy is required for electrons to overcome work function ϕ (1)
Electrons deeper into/below the surface of the metal require more energy to overcome work function and therefore have less KE(1)
Maximum KE = Energy of photon - work function ϕ (1)

Explain how (bright ring) diagram is evidence of the wave-particle duality of electrons? -
ANSWER Diffraction pattern maxima/minima (bright and dark areas) (1)
Diffraction is a property of waves- proves wave-behaviour of electrons (1)

What is meant by the ground state? - ANSWER The lowest energy state of an atom

Explain the difference between excitation and ionisation (2) - ANSWER An electron receives a discrete amount of energy for BOTH (1)
Excitation promotes an electron to a higher energy level
Ionisation is when an electron receives enough energy to leave the atom

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J Rink



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Table of Contents June 2013 Physics Additional Paper Aqa

1. Understanding the eBook June 2013 Physics Additional Paper Aqa
 - The Rise of Digital Reading June 2013 Physics Additional Paper Aqa
 - Advantages of eBooks Over Traditional Books
2. Identifying June 2013 Physics Additional Paper Aqa
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an June 2013 Physics Additional Paper Aqa
 - User-Friendly Interface
4. Exploring eBook Recommendations from June 2013 Physics Additional Paper Aqa
 - Personalized Recommendations

- June 2013 Physics Additional Paper Aqa User Reviews and Ratings
- June 2013 Physics Additional Paper Aqa and Bestseller Lists
- 5. Accessing June 2013 Physics Additional Paper Aqa Free and Paid eBooks
 - June 2013 Physics Additional Paper Aqa Public Domain eBooks
 - June 2013 Physics Additional Paper Aqa eBook Subscription Services
 - June 2013 Physics Additional Paper Aqa Budget-Friendly Options
- 6. Navigating June 2013 Physics Additional Paper Aqa eBook Formats
 - ePub, PDF, MOBI, and More
 - June 2013 Physics Additional Paper Aqa Compatibility with Devices
 - June 2013 Physics Additional Paper Aqa Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of June 2013 Physics Additional Paper Aqa
 - Highlighting and Note-Taking June 2013 Physics Additional Paper Aqa
 - Interactive Elements June 2013 Physics Additional Paper Aqa
- 8. Staying Engaged with June 2013 Physics Additional Paper Aqa
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers June 2013 Physics Additional Paper Aqa
- 9. Balancing eBooks and Physical Books June 2013 Physics Additional Paper Aqa
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection June 2013 Physics Additional Paper Aqa
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine June 2013 Physics Additional Paper Aqa
 - Setting Reading Goals June 2013 Physics Additional Paper Aqa
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of June 2013 Physics Additional Paper Aqa
 - Fact-Checking eBook Content of June 2013 Physics Additional Paper Aqa

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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