

Name _____

Date _____

Period _____

VIDEO GUIDE: NOVA – “Hunting the Elements”

PERIODIC TABLE OF THE ELEMENTS

- 1) How many different elements are there? _____
- 2) What is the symbol for gold? _____
- 3) What is gold's atomic number? _____
- 4) What are some unique properties of gold? _____
- 5) What process is used to separate gold atoms from crushed rock? _____
- 6) How much gold is harvested per ton of earth from the underground mine? _____



7) Gold does not share its electrons easily. Therefore, it is not very chemically reactive. What are two other metals that, like gold, don't react with other elements? _____

8) What is the worth (\$\$) of one 60-pound bar of gold? _____

9) Two rows above gold is the metal: _____; its symbol is: _____

10) In what objects / industries is copper used? _____

11) What are some unique properties of copper? _____

12) What is the symbol for tin? _____ What is its atomic number? _____

13) When copper and tin are mixed together, the alloy _____ is formed.

14) Why isn't copper alone used to make bells? _____



15) It is possible that the crack in the famous Liberty Bell was caused by too much _____

16) Metals form crystals, which means that they have an _____ of atoms.

17) Atoms with 1 proton are the element: _____; atoms with 2 protons are _____.

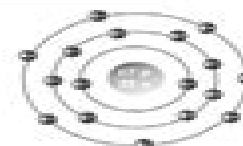
18) The number of protons in an atom is known as that element's _____.

19) The chemical symbol of each element is based on the element's name in which language? _____

20) More than 70% of the elements on the periodic table are _____.

21) Why are the noble gases called "noble"? _____

22) Which subatomic particles determine an element's reactivity? (circle correct answer)

PROTONS**NEUTRONS****ELECTRONS**

23) How many electrons are in the first shell of electrons? _____ The second? _____

24) How many electrons do the noble gases have in their outermost shell? _____

25) How many electrons does chlorine have in its outermost shell? _____

26) What will an atom of chlorine do to complete its outer shell? _____

27) The alkali metals, such as sodium, have _____ electron in their outer shell.

28) What do the alkali metals do with that one electron? _____



Hunting The Elements Worksheet Answers

Léo Pomar



Hunting The Elements Worksheet Answers:

MAX Teaching with Reading and Writing Mark A. Forget, 2007-05-23 Research proven activities that engage students in active processing of new information leading to deeper understanding long term retention of subject matter and acquisition of life long learning skills Intrigue of the Past, 2006 Project Archaeology in Nevada is sponsored by the U S Department of the Interior Bureau of Land Management as well as other public and private partners The goal is to foster stewardship of Nevada s cultural resources and to promote the educational cultural and scientific awareness to benefit diverse present and future generations By introducing the Project Archaeology program into Nevada we hope to equip Nevada s children with the knowledge to make wise decisions regarding the use and preservation of archaeological resources and an appreciation of Nevada s diverse cultural heritage Intro **From Standards to Success** Mark R. O'Shea, 2005-07-15 In this era of accountability and high stakes testing school leaders must find more sophisticated ways to help all students succeed But how can districts make adequate yearly progress without a coherent system for addressing state standards In From Standards to Success education professor Mark R O Shea introduces the Standards Achievement Planning Cycle SAPC a comprehensive protocol for meeting the standards To illustrate his multi layered approach O Shea takes readers to a fictional school as it prepares to install the SAPC We meet the superintendent who organizes the district for curriculum reform the principal who supervises standards based instruction and the teachers who collaboratively plan lessons and evaluate their students work From teacher observation to student assessment O Shea offers innovative strategies to help school leaders identify and analyze which standards are most important select appropriate curriculum materials and resources provide instructional planning time for teachers create a benchmark testing program design effective professional development Checklists at the end of each chapter highlight best practices and sample lessons show how to plan curriculum that enables students to meet state standards The result is a thorough and sensible guide to realizing the promise of standards based education

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