

Name: _____ Date: _____ Period: _____

NOVA: Hunting the Elements

Part 1: Basic Chemistry

Blank Periodic Table

Fill in the parts of this periodic table covered by the documentary.

1 H 1.00794																		2 He 4.002602																			
3 Li 6.941		4 Be 9.012182																		5 B 10.811		6 C 12.011		7 N 14.00643		8 O 15.999		9 F 18.998403		10 Ne 20.1797							
11 Na 22.98976928		12 Mg 24.304																		13 Al 26.9815386		14 Si 28.08558		15 P 30.973762		16 S 32.06		17 Cl 35.453		18 Ar 39.9623831							
19 K 39.0983		20 Ca 40.078		21 Sc 44.955912		22 Ti 47.88		23 V 50.9415		24 Cr 51.9961		25 Mn 54.938045		26 Fe 55.845		27 Co 58.933195		28 Ni 58.6934		29 Cu 63.546		30 Zn 65.38		31 Ga 69.723		32 Ge 72.64		33 As 74.9216		34 Se 78.96		35 Br 79.904		36 Kr 83.80			
37 Rb 85.4678		38 Sr 87.62		39 Y 88.90584		40 Zr 91.224		41 Nb 92.90638		42 Mo 95.94		43 Tc 98		44 Ru 101.07		45 Rh 102.9055		46 Pd 106.3676		47 Ag 107.8642		48 Cd 112.411		49 In 114.818		50 Sn 118.710		51 Sb 121.757		52 Te 127.6		53 I 126.905		54 Xe 131.29			
55 Cs 132.90545196		56 Ba 137.327		57-70 *		71 Lu 174.967		72 Hf 178.49		73 Ta 180.9479		74 W 183.84		75 Re 186.207		76 Os 190.23		77 Ir 192.222		78 Pt 195.084		79 Au 196.966569		80 Hg 200.59		81 Tl 204.3833		82 Pb 207.2		83 Bi 208.9804		84 Po 209		85 At 210		86 Rn 222	
87 Fr 223		88 Ra 226		89-102 (*) (*)		103 Lr 261		104 Rf 261		105 Db 262		106 Sg 266		107 Bh 264		108 Hs 277		109 Mt 268		110 Uun 288		111 Uuu 289		112 Uub 285		114 Uuq											

^a Lanthanide series.

57 La	58 Ce	59 Pr	60 Nd	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb
89 Ac	90 Th	91 Pa	92 U	93 Np	94 Pu	95 Am	96 Cm	97 Bk	98 Cf	99 Es	100 Fm	101 Md	102 No

"Active person"

Gold - Au

1. Write the number of subatomic particles in gold:
 - a. Protons:
 - b. Neutrons:
 - c. Electrons:
2. Give one property of gold.
3. How much gold is in one ton of the mined rock?

Hunting The Elements Worksheet Answer Key

Puja Mehta



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Table of Contents Hunting The Elements Worksheet Answer Key

1. Understanding the eBook Hunting The Elements Worksheet Answer Key
 - The Rise of Digital Reading Hunting The Elements Worksheet Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Hunting The Elements Worksheet Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Hunting The Elements Worksheet Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Hunting The Elements Worksheet Answer Key
 - Personalized Recommendations
 - Hunting The Elements Worksheet Answer Key User Reviews and Ratings
 - Hunting The Elements Worksheet Answer Key and Bestseller Lists
5. Accessing Hunting The Elements Worksheet Answer Key Free and Paid eBooks
 - Hunting The Elements Worksheet Answer Key Public Domain eBooks
 - Hunting The Elements Worksheet Answer Key eBook Subscription Services
 - Hunting The Elements Worksheet Answer Key Budget-Friendly Options

6. Navigating Hunting The Elements Worksheet Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Hunting The Elements Worksheet Answer Key Compatibility with Devices
 - Hunting The Elements Worksheet Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Hunting The Elements Worksheet Answer Key
 - Highlighting and Note-Taking Hunting The Elements Worksheet Answer Key
 - Interactive Elements Hunting The Elements Worksheet Answer Key
8. Staying Engaged with Hunting The Elements Worksheet Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Hunting The Elements Worksheet Answer Key
9. Balancing eBooks and Physical Books Hunting The Elements Worksheet Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Hunting The Elements Worksheet Answer Key
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Hunting The Elements Worksheet Answer Key
 - Setting Reading Goals Hunting The Elements Worksheet Answer Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Hunting The Elements Worksheet Answer Key
 - Fact-Checking eBook Content of Hunting The Elements Worksheet Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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