

# ANGLE PROOFS

**Directions:** Complete the proofs below by giving the missing statements and reasons.

- ① **Given:**  $\angle PQR$  is a right angle  
**Prove:**  $\angle PQS$  and  $\angle SQR$  are complementary



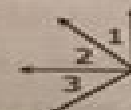
| Statements                                         | Reasons                        |
|----------------------------------------------------|--------------------------------|
| 1. $\angle PQR$ is a right angle                   | 1. Given                       |
| 2. $m\angle PQR = 90^\circ$                        | 2. Def. of complementary angle |
| 3. $m\angle PQS + m\angle SQR = m\angle PQR$       | 3. Transitive property         |
| 4. $m\angle PQS + m\angle SQR = 90^\circ$          | 4. Angle addition Postulate    |
| 5. $\angle PQS$ and $\angle SQR$ are complementary | 5. Def. of complementary angle |

- ② **Given:**  $\angle 2 \cong \angle 3$ ;  $\angle 1$  and  $\angle 2$  form a linear pair  
**Prove:**  $\angle 1$  and  $\angle 3$  are supplementary



| Statements                                      | Reasons  |
|-------------------------------------------------|----------|
| 1. $\angle 2 \cong \angle 3$                    | 1. Given |
| 2. $m\angle 2 = m\angle 3$                      | 2.       |
| 3. $\angle 1$ and $\angle 2$ form a linear pair | 3.       |
| 4. $\angle 1$ and $\angle 2$ are supplementary  | 4.       |
| 5. $m\angle 1 + m\angle 2 = 180^\circ$          | 5.       |
| 6. $m\angle 1 + m\angle 3 = 180^\circ$          | 6.       |
| 7. $\angle 1$ and $\angle 3$ are supplementary  | 7.       |

- ③ **Given:**  $\angle 1$  and  $\angle 2$  form a right angle;  $m\angle 1 + m\angle 3 = 90^\circ$   
**Prove:**  $\angle 2 \cong \angle 3$



| Statements                                      | Reasons  |
|-------------------------------------------------|----------|
| 1. $\angle 1$ and $\angle 2$ form a right angle | 1. Given |
| 2. $\angle 1$ and $\angle 2$ are complementary  | 2.       |
| 3. $m\angle 1 + m\angle 3 = 90^\circ$           | 3.       |
| 4. $\angle 1$ and $\angle 3$ are complementary  | 4.       |
| 5. $\angle 2 \cong \angle 3$                    | 5.       |

# Geometry Unit 8 Answer Key

**JG Myers**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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using basic and advanced features Additional techniques include the edit and reuse of features parts and assemblies through symmetry patterns configurations SOLIDWORKS 3D ContentCentral and the SOLIDWORKS Toolbox Project 7 Understand Top Down assembly modeling and Sheet Metal parts Develop components In Context with InPlace Mates along with the ability to import parts using the Top Down assembly method Convert a solid part into a Sheet Metal part and insert and apply various Sheet Metal features Project 8 Project 9 Recognize SOLIDWORKS Simulation and Intelligent Modeling techniques Understand a general overview of SOLIDWORKS Simulation and the type of questions that are on the SOLIDWORKS Simulation Associate Finite Element Analysis CSWSA FEA exam Apply design intent and intelligent modeling techniques in a sketch feature part plane assembly and drawing Project 10 Comprehend the differences between additive and subtractive manufacturing Understand 3D printer terminology along with a working knowledge of preparing saving and printing CAD models on a low cost printer Project 11 Review the Certified SOLIDWORKS Associate CSWA program Understand the curriculum and categories of the CSWA exam and the required model knowledge needed to successfully take the exam The author developed the industry scenarios by combining his own industry experience with the knowledge of engineers department managers vendors and manufacturers These professionals are directly involved with SOLIDWORKS every day Their responsibilities go far beyond the creation of just a 3D model

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web francis bacon 28 october 1909 28 april 1992 was an irish born british 1 figurative painter known for his raw unsettling imagery focusing on the human form his subjects included crucifixions portraits of popes self portraits and portraits of close friends with abstracted figures sometimes isolated in geometrical structures 2

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terror the son of a racehorse trainer bacon was educated mostly by private tutors at home until his parents banished him at age 16 allegedly for pursuing his homosexual leanings

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concerned

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web objective to apply stoichiometric principles to calculate the appropriate chemical equation for the reaction between iron metal and a solution of copper i sulfate introduction in the form of finely split powder copper metal is produced as a byproduct of the process as iron ions the iron metal dissolves

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web purpose to enhance the understanding of stoichiometry a reaction between iron and copper ii sulfate solution will be conducted this will help you to differentiate limiting and excess reactant in a chemical reaction finally the theoretical and percent yield of this reaction will be calculated

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web october 3 2017 abstract the techniques used in this lab are quantitative transfer and vacuum filtration with the reaction of 8 001 grams of copper ii sulfate  $\text{CuSO}_4$  and 2 0153 grams of iron powder  $\text{Fe}$  the goal of this experiment was to determine the product of copper ii sulfate with iron

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web procedure 1 find the mass of a clean dry 100 ml beaker and record 2 measure out 8 00 grams of copper ii sulfate onto a

piece of filter paper taking the filter paper's mass into account add these to the beaker 3 measure 50.0 ml of water in the cylinder and add it to the beaker 4

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web stoichiometry lab the reaction of iron with copper ii sulfate the study of stoichiometry deals with the calculation of quantities in a chemical reaction how much product will be produced how much reactant do you need to make that much product

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web description great lab for use during stoichiometry unit in chemistry reinforces ideas of mole ratios and stoichiometry calculations by measuring the mass of iron that reacts and the mass of copper metal produced students can calculate the ratio of moles of reactant to moles of product this experimental mole ratio from lab can be compared

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web in this experiment we will use stoichiometric principles to deduce the appropriate equation for the reaction between metallic iron and a solution of copper ii sulfate this reaction produces metallic copper which is seen precipitating as a finely divided red powder

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web in this lab we will use experimental evidence and the principles of stoichiometry to determine which ion is formed in a single replacement reaction between iron and copper ii sulfate pre lab assignment in preparation for the lab activity we will investigate another single replacement reaction between copper and silver nitrate

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web through this experiment we sought to use stoichiometric principles which derive from stoichiometry the study of substance quantities in ratio or proportion in a chemical reaction in this lab experiment the principles help distinguish the equation between iron powder and aqueous copper ii sulfate in the iron copper ii sulfate redox reaction

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web this experiment is focused on a reaction between iron metal and a solution of copper ii sulfate we are able to figure out the chemical equation between the two by using stoichiometric principles the iron displaces the copper ii sulfate ions the copper ii is then reduced and gains electrons

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web stoichiometry can be defined as the relationship between the relative quantities of substances taking part in a reaction or forming a compound stoichiometric principles can be used to deduce the appropriate chemical equation for a reaction in our experiment we used solid iron in the form of powder was added to a solution of copper ii sulfate

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web stoichiometric principles can be used to determine the chemical equation between iron powder and aqueous copper ii sulfate in the iron copper ii sulfate redox reaction the iron ions dissolve in the copper ii sulfate solution to produce copper ii ions

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web under the finalize of the experiment it was found that iron was the limiting reactant and iron ii sulfate was a product of the reaction between 7 0002g away cooper ii sulfate and 2 0101g iron metal which produced 2 3037g of copper product with a percent yield of 100 72 introduction

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web stoichiometry lab report the reaction of iron with copper ii sulfate period 5 in this experiment i observed the reaction of iron with copper ii sulfate i obtained the results to this experiment through adding heated  $\text{CuSO}_4$  solution to the iron powder stirring the solution adding water then removing water adding acetone heating the acetone on low

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web stoichiometry the reaction of iron with copper ii sulfate purpose to enhance the understanding of stoichiometry a reaction between iron and copper ii sulfate solution will be conducted this will help you to differentiate limiting and excess reactant in a chemical reaction

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web stoichiometry lab iron and copper sulfate sample data data and observations mass of copper ii sulfate pentahydrate 3 27 g of g mass of filter paper mass of filter paper and dry copper mass of dry copper 1 28 g

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