

# Types of Nuclear Radiation – Questions

**Instructions:** Identify the following descriptions as alpha, beta, or gamma:

- |  |              |
|--|--------------|
| 1) This radiation can only be stopped by <u>thick concrete or lead</u> .   | <u>Gamma</u> |
| 2) This particle can be stopped by <u>clothing alone</u> .                 | <u>Alpha</u> |
| 3) During this decay, <u>no protons, neutrons, or electrons are lost</u> . | <u>Gamma</u> |
| 4) During this type of decay, <u>two protons</u> are lost.                 | <u>Alpha</u> |
| 5) This particle has a <u>-1 charge</u> .                                  | <u>Beta</u>  |
| 6) This particle has a mass of 4.  | <u>Alpha</u> |
| 7) This type of radiation is identical to an <u>electron</u> .             | <u>Beta</u>  |
| 8) This form of radiation has <u>no mass</u> and <u>no charge</u> .        | <u>Gamma</u> |
| 9) This particle is represented as a <u>helium</u> atom.                   | <u>Alpha</u> |
| 10) This particle is emitted as in the form of waves.                      | <u>Gamma</u> |
| 11) This particle can be stopped by a <u>piece of paper</u> .              | <u>Alpha</u> |
| 12) This particle has average (medium) penetrating power.                  | <u>Beta</u>  |
| 13) This is the <u>least dangerous</u> type of radiation.                  | <u>Alpha</u> |
| 14) This is the <u>most dangerous</u> type of radiation.                   | <u>Gamma</u> |

**Instructions:** Use the nuclear equation below to answer the following questions:



- |   |                    |
|---|--------------------|
| 15) The picture above represents <u>what type</u> of nuclear decay? | <u>Beta Decay</u>  |
| 16) What is the mass of the starting phosphorus (P) atom?           | <u>32</u>          |
| 17) What is the mass of the particle released above?                | <u>0</u>           |
| 18) Which decay would have caused a larger change in mass?          | <u>Alpha Decay</u> |

# Key Terms Radioactivity And Nuclear Reactions Worksheet Answer

**Hugh D. Young, Roger A. Freedman**



## **Key Terms Radioactivity And Nuclear Reactions Worksheet Answer:**

Disaster Preparedness Specialist (AFSC 24250) Joseph L. Glenn, 1984      *General Chemistry for Engineers* Jeffrey Gaffney, Nancy Marley, 2017-11-13 *General Chemistry for Engineers* explores the key areas of chemistry needed for engineers. This book develops material from the basics to more advanced areas in a systematic fashion. As the material is presented, case studies relevant to engineering are included that demonstrate the strong link between chemistry and the various areas of engineering. Serves as a unique chemistry reference source for professional engineers. Provides the chemistry principles required by various engineering disciplines. Begins with an atoms first approach building from the simple to the more complex chemical concepts. Includes engineering case studies connecting chemical principles to solving actual engineering problems. Links chemistry to contemporary issues related to the interface between chemistry and engineering practices.

**Introductory Chemistry** Charles H. Corwin, 2005 For one semester courses in Basic Chemistry. Introduction to Chemistry and Preparatory Chemistry and the first term of Allied Health Chemistry. This text is carefully crafted to help students learn chemical skills and concepts more effectively. Corwin covers math and problem solving early in the text; he builds student confidence and skills through innovative problem solving pedagogy and technology formulated to meet student needs.      *Understanding Chemistry* Fred M. Dewey, 1994      *Regents Exams and Answers: Chemistry--Physical Setting Revised Edition* Barron's Educational Series, Albert Tarendash, 2021-01-05 Barron's *Regents Exams and Answers: Chemistry* provides essential practice for students taking the Chemistry Regents, including actual recently administered exams and thorough answer explanations for all questions. This book features eight actual administered Regents Chemistry exams so students can get familiar with the test. Thorough explanations for all answers. Self-analysis charts to help identify strengths and weaknesses. Test-taking techniques and strategies. A detailed outline of all major topics tested on this exam. A glossary of important terms to know for test day.      University Physics Hugh D. Young, Roger A. Freedman, 2000 This work aims to enable students to develop physical intuition and strong problem-solving skills. In addition, it points out the conceptual and computational pitfalls that commonly plague beginner physics students.      **Chemical Matter** Prentice-Hall Staff, 1994 Authorized teaching resource in Alberta for senior high science 14/24 1995/2004      Chemistry & Chemical Reactivity John C. Kotz, Paul Treichel, Patrick A. Harman, 2003 Provides a broad overview of the principles of chemistry, the reactivity of chemical elements and their compounds, and the applications of chemistry. Conveys a sense of chemistry as a field that not only has a lively history but also one that is currently dynamic with important new developments on the horizon.

*Introduction to Physical Science* James T. Shipman, 2000-06      **Chemistry** Karen Timberlake, 2003 *Chemistry: An Introduction to General Organic and Biological Chemistry* now in its eighth edition makes chemistry exciting by showing why important concepts are relevant to the lives and future careers of readers. The new design, digital images, photos, Career Focus features, and macro to micro art enhance the new edition while it retains the many features that have made this book

so successful The writing as always is exceptionally friendly Each section contains sample problems that develop readers critical thinking skills This edition also contains more conceptual problems than ever before and has been redesigned to accomodate new styles of learning and teaching with a wide variety of pedagogical tools Health and Environmental Notes throughout the book highlight topics that are relevant to readers lives and are ideal for classroom discussion Explore Your World activities in each chapter make chemistry exciting relevant and non threatening

An Introduction to Physical Science James T. Shipman, Jerry D. Wilson, Aaron W. Todd, 2002-07-08 This text presents a survey of the physical sciences physics chemistry astronomy meteorology and geology for non science majors Topics are treated both descriptively and quantitatively providing flexibility for instructors who wish to emphasize a highly descriptive approach a highly quantitative approach or any spectrum in between

Introduction to Chemical Principles Howard Stephen Stoker, 1999 Chemistry for students who need full exposure to general chemistry but in compact one semester 17 chapter paperback format Strong emphasis on problem solving with over 5000 problems in end of chapter material arranged in matched pairs More real life applications added to this edition plus faces of chemistry

Fundamentals of Chemistry Elizabeth P. Rogers, 1987

Prentice Hall Chemistry, 2000

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*Optics and Modern Physics for JEE Advanced, 3E (Free Sample)* B. M. Sharma, 2022-05-19 Optics and Modern Physics for JEE Advanced a Cengage Exam Crack Series product is designed to help aspiring engineers focus on the subject of physics from two standpoints To develop their caliber aptitude and attitude for the engineering field and profession To strengthen their grasp and understanding of the concepts of the subjects of study and their applicability at the grassroots level Each book in this series approaches the subject in a very conceptual and coherent manner While its illustrative solved examples facilitate easy mastering of the concepts and their applications an array of solved problems exposes the students to a variety of questions that they can expect in the examination The coverage and features of this series of books make it highly useful for all those preparing for JEE Main and Advanced and aspiring to become engineers

*Foundations of College Chemistry* Morris Hein, Susan Arena, Cary Willard, 2016-08-02 This text is an unbound three hole punched version Used by over 750 000 students

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*Preview Edition for Chemistry* Moore, 2001-03

*Introduction to Modern Chemistry:*

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