## Question 4

4(a) Radiocarbon dating using <sup>16</sup>C is a method of determining the age of once living, archaeological specimens which are no more than 50 000 years old.

123 is an example of a substance which can be used as a tracer. Tracers which concentrate in some organs are introduced into the bodies of living organisms and used to study chemical changes.

4(b)(i) Number of occupied electron shells = period number. Number of electrons in outer shell = group number.

	P	Q.	R	S
Group #	7	2	6	2
Period#	2	3	3	4

4(b)(ii) The compound is ionic.



Formula is QP2.

4(b)(iii) S will react more vigorously with hydrochloric acid than Q because S is lower in group 2 so it loses its outer electrons more easily than Q.

Equation for reaction of Q with HCI:  $Q_{(s)} + 2HCI_{(aq)} \rightarrow QCI_{2(aq)} + H_{2(g)}$ .

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## Table of Contents F32chemistry June 2014 Paper

- 1. Understanding the eBook F32chemistry June 2014 Paper
  - The Rise of Digital Reading F32chemistry June 2014 Paper
  - Advantages of eBooks Over Traditional Books
- 2. Identifying F32chemistry June 2014 Paper
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an F32chemistry June 2014 Paper
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from F32chemistry June 2014 Paper
  - Personalized Recommendations
  - F32chemistry June 2014 Paper User Reviews and Ratings
  - F32chemistry June 2014 Paper and Bestseller Lists

- 5. Accessing F32chemistry June 2014 Paper Free and Paid eBooks
  - F32chemistry June 2014 Paper Public Domain eBooks
  - F32chemistry June 2014 Paper eBook Subscription Services
  - F32chemistry June 2014 Paper Budget-Friendly Options
- 6. Navigating F32chemistry June 2014 Paper eBook Formats
  - ∘ ePub, PDF, MOBI, and More
  - F32chemistry June 2014 Paper Compatibility with Devices
  - F32chemistry June 2014 Paper Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of F32chemistry June 2014 Paper
  - Highlighting and Note-Taking F32chemistry June 2014 Paper
  - Interactive Elements F32chemistry June 2014 Paper
- 8. Staying Engaged with F32chemistry June 2014 Paper
  - o Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers F32chemistry June 2014 Paper
- 9. Balancing eBooks and Physical Books F32chemistry June 2014 Paper
  - $\circ$  Benefits of a Digital Library
  - $\circ\,$  Creating a Diverse Reading Collection F32chemistry June 2014 Paper
- 10. Overcoming Reading Challenges
  - $\circ\,$  Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine F32chemistry June 2014 Paper
  - Setting Reading Goals F32chemistry June 2014 Paper
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of F32chemistry June 2014 Paper
  - Fact-Checking eBook Content of F32chemistry June 2014 Paper
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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