F321: Atoms, Bonds and Groups Redox

39 Marks

 The element strontium forms a nitrate, Sr(NO₃)₂, which decomposes on heating as shown below.

 $2Sr(NO_3)_2(s) \rightarrow 2SrO(s) + 4NO_2(g) + O_2(g)$

- (i) Using exidation numbers, explain why the reaction involves both exidation and reduction.
- (ii) A student heats 5.29 g of Sr(NO₃)₂ and collects the gas at room temperature and pressure, RTP.

$$2Sr(NO_3)_2(s) \rightarrow 2SrO(s) + 4NO_2(g) + O_2(g)$$

Calculate the volume of gas, in \mbox{dm}^3 , obtained by the student at RTP.

Molar mass of $Sr(NO_3)_2 = 211.6 g mol^{-1}$.

[3]

[Total 6 marks]

121

F321 Atoms Bonds And Groups June 2014

Arturo Cuomo



The Enigmatic Realm of F321 Atoms Bonds And Groups June 2014: Unleashing the Language is Inner Magic

In a fast-paced digital era where connections and knowledge intertwine, the enigmatic realm of language reveals its inherent magic. Its capacity to stir emotions, ignite contemplation, and catalyze profound transformations is nothing short of extraordinary. Within the captivating pages of **F321 Atoms Bonds And Groups June 2014** a literary masterpiece penned by a renowned author, readers attempt a transformative journey, unlocking the secrets and untapped potential embedded within each word. In this evaluation, we shall explore the book is core themes, assess its distinct writing style, and delve into its lasting affect the hearts and minds of people who partake in its reading experience.

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