Chapter

7

THERMOCHEMISTRY

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The second second

MULTIPLE CHOICE QUESTIONS

1.	Spen	taneous reacti	ions are	er.					CLHIR 05)	
	(40)	Reversible	(b)	Irreversible	(c)	Mon I	rneversible	(4)	None of these	
2.	The standard heat changes occur at: (BWP 08, GRW 06)									
	(a)	25°C and 2 at				(b)	298 K and 1 a			
	(c) 25°C and 1 mm hg					(d)	273 K and 1 at	mo.		
3.	H" + OH"→ H ₂ O the change in enthalpy for reaction is called: (UHROS)									
	(a)	Heat of reacti				(b)	heat of forms			
	(c)	Heat of neutri				(4)	Heat of comb			
4.				y in a chemica	al react	ion is			es place directly or	
		ectly. It is call							1, GRW 07, BWP 10)	
	(40)	Henry's law	(b)	Charlie's law	. Allen and a second	(c)		(d)	Graham's law	
5.	The enthalpy change when one mole of substance is completely burnt in excess of oxygen is									
	called: (LHR 07)									
	(a) Enthalpy of atomization					(P)	Enthalpy of neutralization			
	(c) Enthalpy of Combustion					(d)	Enthalpy of formation			
6.	Enthalpy change for the reaction: $CH_{4(n)} + 2O_{2(n)} \rightarrow CO_{2(n)} + 2H_2O_{(0)}$ is called enthalpy of: $(LH_2O_{(0)})$									
	(a)	Formation	(b)	Combustion		(c)	Neutralization	(d)	Atomization	
7.	Standard enthalpies are measured at: (ESO 11, LHR 09)									
	(a)	273 K	(b)	2:9BK		(<)	373 K	(4)	All of these	
8.								IK 10, GRW, LHR 11)		
	(a)	Evaporation	(b)	Sublimation	and the contract	(c)	Respiration	(d)	Boiling	
9.	A state function which describes together the internal energy and the product of pressure and									
	volume is called:							(GRW 11)		
	(a)	Enthalpy	(b)	Internal energ	PY:	(c)	Wienk	(d)	Free energy	

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