

Book 3: Example questions. All questions were translated from German. Keys are marked with an asterisk [*]. See text for details.

Question 1. Example of a kypriase-type question.

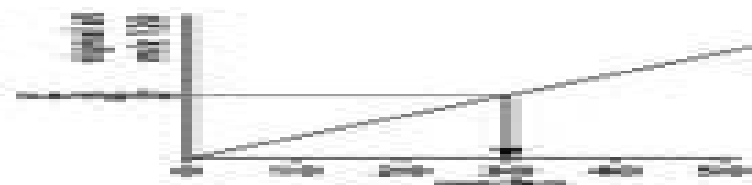
What increases the spectral resolution of an ICP-MS instrument?

- | | | |
|-----------------------|-----------------------|---|
| True | False | |
| ☐ | ☐ | Scatter plate in a sector field instrument |
| ☐ | ☐ | Using a dual mode detector with analog and pulse counting |
| ☐ | ☐ | Detection of barium at m/z 138 instead of 137 |
| ☐ | ☐ | Increase the length of the quadrupole for an ICP-QMS |

Question 2. Silver has a mass of 107.87 g/mol. At which m/z do you detect silver with ICP-MS?

- | | |
|----|-----|
| A | 106 |
| B* | 107 |
| C | 108 |
| D* | 109 |

Question 3. You quantify tin in a sample with ICP-MS and external calibration at m/z 120. The isotope abundance of ^{118}Sn is about 30%. The calibration for the signal at m/z 120 indicates a mass fraction of 30 $\mu\text{g}/\text{kg}$ [see figure below]. What is the mass fraction of tin in the measurement aliquot?



- | | |
|----|----------------------------|
| A | 90 $\mu\text{g}/\text{kg}$ |
| B | 60 $\mu\text{g}/\text{kg}$ |
| C* | 30 $\mu\text{g}/\text{kg}$ |
| D | 10 $\mu\text{g}/\text{kg}$ |

Question 4. Chocolate was digested in a closed vial under pressure using nitric acid and hydrogen peroxide using a microwave system. What constitutes the main matrix of the analyte cadmium in the prepared solution after the digestion of chocolate?

- | | |
|----|-------------------|
| A | cadm., sugar, fat |
| B* | water |
| C | cadmium ions |
| D | no matrix |

Question 5. How do you prepare a sample of selenium to differentiate between and quantify organic mercury species?

- | | |
|----|---|
| A | microwave digestion using HCl and H_2O_2 |
| B | microwave digestion using HNO_3 and H_2O_2 |
| C | microwave digestion using HNO_3 |
| D* | no digestion |

Multiple Choice Question For Instrumental Chemistry

Marcel A. Müller



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