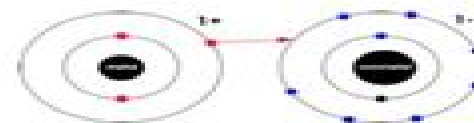


DRAWING IONIC BONDS

Introduction:

An ionic bond forms between a metal and a nonmetal. An ionic bond occurs when a metal atom **loses valence electron(s)** to become a positive (+) ion or **cation**, while the nonmetal **gains valence electron(s)** to become a negative (-) ion or **anion**. An ionic bond forms due to the strong electrostatic attraction between the **positive and negative ions**.



How to Draw Simple Ionic Bond Diagrams

1. Draw the electron dot diagrams for each element.

***TIP:** This will help you see the number of unpaired valence electrons for each element, which will help you determine how many electrons each element must gain or lose to achieve a full outer shell.*

2. Draw arrows to show the transfer of valence electrons from the metal to the nonmetal.

WORKSHEET



Drawing Ionic & Covalent Bonds 2 Worksheet Set

	Metal	Nonmetal	Formula	Bonding Diagram
A.	sodium <i>(must lose 1)</i> $\text{Na} \cdot$	chlorine <i>(must gain 1)</i> $\cdot \ddot{\text{Cl}} \cdot$	NaCl <i>(shows 1 Na and 1 Cl)</i>	
B.	calcium <i>(must lose 2)</i> $\cdot \text{Ca} \cdot$	fluorine <i>(must gain 1)</i> $\cdot \ddot{\text{F}} \cdot$	CaF_2 <i>(shows 1 Ca and 2 F)</i>	

Ionic Covalent Bond Practice Worksheets

Brendan G. Carr

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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