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1)	Choose the one a	memative that	best cox	apoetes the	statement or answers the qu	estion.
Which of the: A) It has 12 B) It has 6; C) Its atom D) It has 6;	following statemer neutrons in its nu- protons in its mucl- ic weight is 12. electrons orbiting ic number is 6.	icleus. eus.	om. ⁶ C	ts FALSE	?	
			Table 2.1	i.		
		16	12	1		
		80	6 C	1 _H		
2) Using the info	rmation in Table	2.1, calculate the	e moleci	alar weight	t of ethanol, C2H5OH.	2)
A) 33						
B) 96						
C) 46						
D) 34						
E) The answ	ver cannot be dete	ermined.				
	ralize acid by the HCl → MgCl ₂ + H		on. Idea	ntify the sa	It in the following equation:	3)
A) H2O						
B) MgCl ₂						
C) HCl						
D) Mg(OH)	2					
E) None of	the answers is cor	mect.				
4) Which of the	following statemer	nts is FALSE?				4)
A) Water freezes from the top down.						
B) Salts rea	dily dissolve in w	ater.				
	olecules are form					
	a part of a dehyd:	ration synthesis	reaction	TL.		
E) Water is	a polar molecule.					
5) Which of the	5) Which of the following is the type of bond holding K+ and I- ions in KI?					
 A) hydroge 	n bond	B) sense be	nd		C) covalent bond	
6) Which of the following is the type of bond between molecules of water in a beaker of water?						6)
A) hydroge	n bond	B) covalent	bond:		C) toruc bond	
7) What is the type of bond holding hydrogen and oxygen atoms in the H2O molecule?						7)
A) hydroge	n bond	B) sense be	nd.		C) covalent bond	
8) Identify the following reaction: Glucose + Fructose → Sucrose + Water						S)
A) reversible	e reaction.					40000
B) exchang	e reaction					
C) delvydra	tion synthesis rea	ction				
D) hwdroly:	ds reaction					

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