

STUDY PLAN for your Kaplan NCLEX-RN® Preparation

Welcome to Kaplan NCLEX® Prep! This is the recommended study plan for utilizing the robust Kaplan NCLEX® Prep resources. These resources are designed to give you plenty of realistic practice for success on NCLEX® Test Day. As you begin your preparation, please keep in mind that preparing for the NCLEX® is a marathon, not a sprint! It is important not to rush your studies, but it's also important not to procrastinate.

While you are utilizing the Kaplan NCLEX® Prep resources, it is important that you follow these steps: analyze, review/remediate, think, study, and continue practicing. Remediation is the key to your NCLEX® success.

Phase 1: Prior to Your Kaplan NCLEX® Review Course

Complete the following resources on your student homepage.

***Please note: Directions in italics apply only to institutional students who are enrolled in the Kaplan Institutional program via their school and who access their NCLEX® resources via their Kaplan Institutional homepage.*

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	Download the NCLEX-RN® Content Review Guide e-book . It can be downloaded to any computer or e-reader device.
	Watch the 5 videos under "Pathway to NCLEX® Success."
	Watch the Phase One orientation video.
	Take the Diagnostic Test . <i>** If you are a Kaplan Institutional student, this will be proctored by your school or you will receive at-home testing instructions approximately 7–10 days prior to your NCLEX® class start. This resource will be located in the Green Integrated Testing box. (All other NCLEX® resources are in the Gray NCLEX® Prep box.) Your school may also choose to replace this test with a proctored Kaplan Secured Predictor exam.</i>
	Take Question Trainer 1 . Review/remediate before progressing to Question Trainer 2, and then follow the steps listed below.
	Take Question Trainer 2 . Review/remediate before progressing to Question Trainer 3, and then follow the steps listed below.
	Take Question Trainer 3 . Review/remediate and follow the steps listed below.
	Watch the Essential Nursing Content Videos .
	Download the Classroom Posters . These will be referenced in the Review Course.

As you are completing tests, look at your Analysis page to see how you performed on each test. Aim for scores of 65% or higher. (Institutional students: your school may require higher scores.)

Complete the **Test Reflection Worksheet** to identify your strengths and weaknesses. Ask yourself the following and think about how you are thinking:

1. Did I change any answers? Which way: correct to incorrect, or incorrect to correct?
2. Did I take enough time on each question, or did I take too much time on each question?
3. Did I lose concentration, and if so, is there a pattern? Should I have taken a break?

Review/remediate all questions. Ask yourself the following:

1. Did I not know the content of any of the questions?
2. For each question I got wrong, why did the author of the question choose one answer and why did I choose another?

Use the **Essential Nursing Content Videos** and your **NCLEX-RN® Content Review Guide e-book** to fill in any knowledge gaps, working from your content area of greatest weakness toward your area of greatest strength.

Kaplan Study Guide P

**John Marx, Robert Hockberger, Ron
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web what does the crossed arrow represent in the comic strip 11 what are two of the definitions of dipole given in the comic strip 12 making sense 13 if you finish early what does electronegativity have to do with polarity using polar bears and penguins create an illustration showing a hydrogen sulfide molecule H_2S

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web each dipole represents one polar bond the molecule is nonpolar 10 describe what you think is happening to the penguins in the illustration of H_2O on page 4 they are being pulled towards the stronger polar bear he's more electronegative 11

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web answer because the two polar bears or penguins have equal electronegativity they will pull electrons equally hard and in most cases neither will end up with more electrons each polar bear and penguin has an electronegativity of the same value a non polar bond is formed when two polar bears or two penguins are bonded together

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web why does the polar bear tell the penguin to try to be more positive the polar bear benefits more when the penguin is more positive the more the polar bear attracts the shared pair of electrons the more positive the partial charge on the penguin

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web answer the following questions consider the following illustration draw the lewis dot structure for hcl if the penguin represents a hydrogen atom and the polar bear represents a chlorine atom what does the ice cream represent in the drawing what do you think the picture is trying to illustrate would hcl be attracted to the charged wand

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web to answer this question you will examine 1 electronegativity 2 nonpolar vs polar molecules 3 electronegativity and bonding electronegativity the h atom and the cl atom in hydrogen chloride

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web 1 how does the comic strip define a polar molecule 2 define electronegativity as you understand it after reading the first two pages of the comic strip 3 what is the artist trying to represent by two polar bears arm wrestling or two penguins arm wrestling 4 what three types of bonds are represented on the third page of the comic strip

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web explain how the iceberg penguins and polar bears represent trends in electronegativity on the periodic table and more study with quizlet and memorize flashcards containing terms like how does the comic define a polar molecule define electronegativity as you understand it after reading the first two pages of the comic interpret the pic

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web so the final representation of the nh₃ molecule using polar bears and penguins is as follows begin array c text polar bear nitrogen text with two lone pairs of electrons text and three penguins hydrogen end array this is a fun and creative way to represent the nh₃ molecule

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