

SECTION - A

Q. 1. Select and write the correct answer for the following multiple choice type of questions :

[10]

(i) The product obtained in the following reaction

P 239
11.2



- (a) Pent-3-en-1-ol ✓(b) Pentan-1-ol
(c) Pentan-2-ol (d) Pentanal

(ii) Amongst the following, the solubility of which ionic solid decreases with increase in temperature?

P 30
11)

- (a) KNO_3 (b) NaBr
✓(c) Na_2SO_4 (d) KCl

(iii) The correct IUPAC name of $\text{Na}_3 [\text{AlF}_6]$ is

P 196
Table
9.3

- ✓(a) Sodium hexafluoroaluminate (III)
(b) Sodium hexafluoroaluminate (II)
(c) Sodium hexafluoroaluminium (III)
(d) Sodium hexafluoroaluminium (II)

(iv) Which of the following acids has highest pKa value?

P 277
Table
12.9

- (a) Mono chloroacetic acid (b) Dichloroacetic acid
(c) Trichloroacetic acid ✓(d) Acetic acid

(v) Number of carbon atoms present in isoprene unit is ____.

P 322
15.3

- (a) 6 ✓(b) 5
(c) 4 (d) 3

(vi) The colourless transition metal ion amongst the following is

P 173

- (a) Cu^+ 3d¹⁰ (b) Cu^{++}
(c) Ni^{++} (d) Co^{++}

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Table of Contents Hsc Unregular Chemistry Question Paper 2bd

1. Understanding the eBook Hsc Unregular Chemistry Question Paper 2bd
 - The Rise of Digital Reading Hsc Unregular Chemistry Question Paper 2bd
 - Advantages of eBooks Over Traditional Books
2. Identifying Hsc Unregular Chemistry Question Paper 2bd
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Hsc Unregular Chemistry Question Paper 2bd
 - User-Friendly Interface
4. Exploring eBook Recommendations from Hsc Unregular Chemistry Question Paper 2bd
 - Personalized Recommendations
 - Hsc Unregular Chemistry Question Paper 2bd User Reviews and Ratings
 - Hsc Unregular Chemistry Question Paper 2bd and Bestseller Lists
5. Accessing Hsc Unregular Chemistry Question Paper 2bd Free and Paid eBooks
 - Hsc Unregular Chemistry Question Paper 2bd Public Domain eBooks
 - Hsc Unregular Chemistry Question Paper 2bd eBook Subscription Services
 - Hsc Unregular Chemistry Question Paper 2bd Budget-Friendly Options
6. Navigating Hsc Unregular Chemistry Question Paper 2bd eBook Formats

- ePub, PDF, MOBI, and More
 - Hsc Unregular Chemistry Question Paper 2bd Compatibility with Devices
 - Hsc Unregular Chemistry Question Paper 2bd Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Hsc Unregular Chemistry Question Paper 2bd
 - Highlighting and Note-Taking Hsc Unregular Chemistry Question Paper 2bd
 - Interactive Elements Hsc Unregular Chemistry Question Paper 2bd
 8. Staying Engaged with Hsc Unregular Chemistry Question Paper 2bd
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Hsc Unregular Chemistry Question Paper 2bd
 9. Balancing eBooks and Physical Books Hsc Unregular Chemistry Question Paper 2bd
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Hsc Unregular Chemistry Question Paper 2bd
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Hsc Unregular Chemistry Question Paper 2bd
 - Setting Reading Goals Hsc Unregular Chemistry Question Paper 2bd
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Hsc Unregular Chemistry Question Paper 2bd
 - Fact-Checking eBook Content of Hsc Unregular Chemistry Question Paper 2bd
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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