# Limiting Reagent

Absence of limiting reagent

$$3 H_2 + N_2 \rightarrow 2 NH_3$$
6 g 28 g 34 g

Presence of limiting reagent

1.5 
$$H_2$$
 +  $N_2$   $\rightarrow$  28 g

Limiting reagent

+ N<sub>2</sub>
14 g remaining
Excess
reagent

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17 g

# **Holt Chemistry Concept Review Limiting Reactants Problems**

**RJ Alexander** 

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