

Chapter Test Chapter 10: Matter and Energy

Matter and Energy

Write the letter of the correct answer in the space at the end of the question. Some questions have more than one correct answer. Write all that apply.

1. Which of the following is not a physical property of matter?
 - a. color
 - b. mass
 - c. density
 - d. volume
2. Which of the following is a chemical property?
 - a. the substance is white and has a crystalline structure
 - b. the substance is soluble in water
 - c. the substance is flammable
 - d. the substance is a solid at room temperature
3. Which of the following is a physical property of matter?
 - a. color
 - b. mass
 - c. density
 - d. volume
4. Which of the following is a chemical property of matter?
 - a. the substance is white and has a crystalline structure
 - b. the substance is soluble in water
 - c. the substance is flammable
 - d. the substance is a solid at room temperature
5. Which of the following is a physical property of matter?
 - a. color
 - b. mass
 - c. density
 - d. volume
6. Which of the following is a chemical property of matter?
 - a. the substance is white and has a crystalline structure
 - b. the substance is soluble in water
 - c. the substance is flammable
 - d. the substance is a solid at room temperature
7. Which of the following is a physical property of matter?
 - a. color
 - b. mass
 - c. density
 - d. volume
8. Which of the following is a chemical property of matter?
 - a. the substance is white and has a crystalline structure
 - b. the substance is soluble in water
 - c. the substance is flammable
 - d. the substance is a solid at room temperature

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**John A. Olmsted, Gregory M.
Williams, Robert Charles Burk**



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Modern Electrochemistry 2B John O'M. Bockris, Amulya K.N. Reddy, 2007-05-08 This book had its nucleus in some lectures given by one of us J O M B in a course on electrochemistry to students of energy conversion at the University of Pennsylvania. It was there that he met a number of people trained in chemistry physics biology metallurgy and materials science all of whom wanted to know something about electrochemistry. The concept of writing a book about electrochemistry which could be understood by people with very varied backgrounds was thereby engendered. The lectures were recorded and written up by Dr Klaus Muller as a 293 page manuscript. At a later stage A K N R joined the effort it was decided to make a fresh start and to write a much more comprehensive text. Of methods for direct energy conversion the electrochemical one is the most advanced and seems the most likely to become of considerable practical importance. Thus conversion to electrochemically powered transportation systems appears to be an important step by means of which the difficulties of air pollution and the effects of an increasing concentration in the atmosphere of carbon dioxide may be met. Corrosion is recognized as having an electrochemical basis. The synthesis of nylon now contains an important electrochemical stage. Some central biological mechanisms have been shown to take place by means of electrochemical reactions. A number of American organizations have recently recommended greatly increased activity in training and research in electrochemistry at universities in the United States.

Chemistry John A. Olmsted, Gregory M. Williams, Robert Charles Burk, 2016-01-14 Olmsted Burk is an introductory general chemistry text designed specifically with Canadian professors and students in mind. A reorganized Table of Contents and inclusion of SI units IUPAC standards and Canadian content designed to engage and motivate readers distinguish this text from many of the current text offerings. It more accurately reflects the curriculum of most Canadian institutions. Instructors will find the text sufficiently rigorous while it engages and retains student interest through its accessible language and clear problem solving program without an excess of material that makes most text appear daunting and redundant.

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