

Chapter Test B, continued _____

17. A substance combines with oxygen, releasing a large amount of energy as heat and light, as when _____.
18. The decomposition of a substance by an electric current is called _____.
19. A(n) _____ causes the elements by the mass with which they combine remain chemically unaltered.
20. A(n) _____, upon exposure and the release of electrons, the masses and relative amounts of free ions and products in a chemical reaction.
21. The law of _____ means that not all that is chemical equation to be balanced.
22. A(n) _____ product is sometimes indicated by an asterisk following equation.
23. Two equalled arrows mean that the reactions are _____.

PART II Write the answers to the following questions in the space provided.

24. How are a word equation and a formula equation alike? How are they alike (How is chemical equation)?
- _____
- _____
- _____
25. What are three observations that indicate a chemical change may be taking place?
- _____
- _____
- _____

PART III Write a word equation for each chemical reaction in the space provided.



Holt Chemistry Assessment Test Answers

Susan E. Kegley, Laura Wise



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