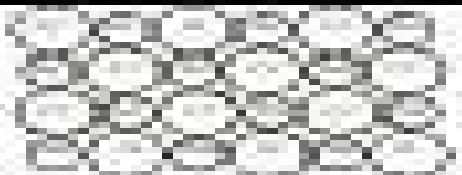
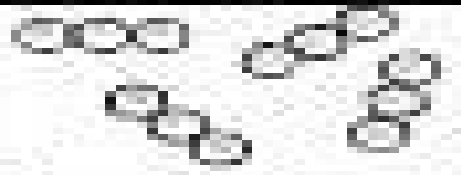
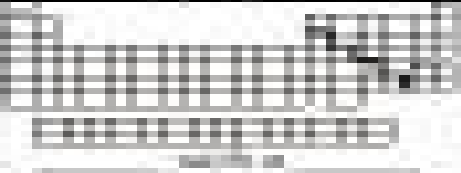
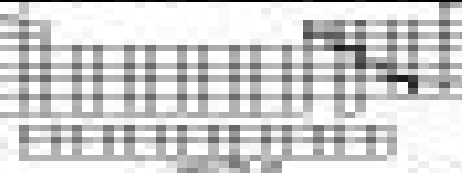
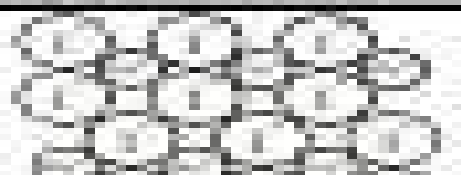
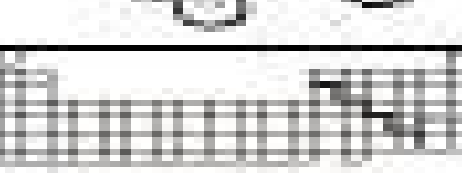
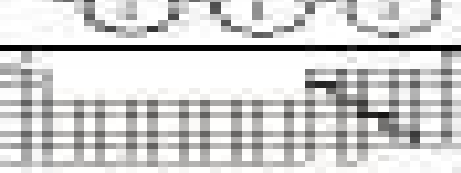


Particle connections – What's in a name?

Object: Illustrate the connection between the particles and the macroscopic properties of a gas. How do the number, position, and energy of the particles relate to the macroscopic properties of a gas?

Figure 1

Particle model			
Macroscopic properties			
Equation	$P = \frac{F}{A}$	$P = \frac{F}{A}$	$P = \frac{F}{A}$
Equation	$P = \frac{F}{A}$	$P = \frac{F}{A}$	$P = \frac{F}{A}$
Equation	$P = \frac{F}{A}$	$P = \frac{F}{A}$	$P = \frac{F}{A}$
Particle model			
Macroscopic properties			
Equation	$P = \frac{F}{A}$	$P = \frac{F}{A}$	$P = \frac{F}{A}$
Equation	$P = \frac{F}{A}$	$P = \frac{F}{A}$	$P = \frac{F}{A}$

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