#### Chapter 5 Practice Test

#### Multiple Choice

Identify the choice that best completes the statement or answers the question.

- 1. The product of the frequency and the wavelength of a wave equals the
  - a. number of waves passing a point in a second.
  - b. speed of the wave.
  - distance between wave crests.
  - d. time for one full wave to pass.
- 2. Visible light, X rays, infrared radiation, and radio waves all have the same
  - a. energy.

c. speed.

b. wavelength.

- d. frequency.
- 3. The speed of an electromagnetic wave is equal to the product of its wavelength and its
  - a. mass.

c. velocity.

b. color.

- d. frequency.
- Because c, the speed of electromagnetic radiation, is a constant, the wavelength of the radiation is
  - a. proportional to its frequency.

c. inversely proportional to its frequency.

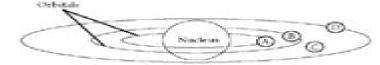
b. equal to its frequency.

- d. double its frequency.
- For an electron in an atom to change from the ground state to an excited state.
  - a. energy must be released.
  - b. energy must be absorbed.
  - c. radiation must be emitted.
  - d. the electron must make a transition from a higher to a lower energy level.
- 6. If electrons in an atom have the lowest possible energies, the atom is in the
  - a. ground state.

c: excited state.

b. inert state.

- d. radiation-emitting state.
- According to Bohr, electrons cannot reside at in the figure below.



- a. point A.
- b. peoint B.

- c. point C
- d. point D
- 8. A spherical electron cloud surrounding an atomic nucleus would best represent a. an x orbital.

  - a ρ, orbital.
  - a combination of p, and p, orbitals.
  - d. a combination of an s and a p, orbital.
- 9. The total number of orbitals that can exist at the second main energy level is

## **Holt Chapter Test Atoms And Periodic Table Answers**

**E Staff** 

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