

# A. Limiting Reactants



<del>0.90 L</del>	<del>2.5 mol HCl</del>	<del>1 mol H<sub>2</sub></del>	22.4 L H <sub>2</sub>	= 25 L H <sub>2</sub>
	<del>1 L</del>	<del>2 mol HCl</del>	<del>1 mol H<sub>2</sub></del>	

# Holt Chemistry Limiting Reactant And Percentage Yield

**Raffaela Di Napoli**



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