

Skills Worksheet

Concept Review Chapter 5: Section 1

Section: Simple Ions

Complete each statement below by choosing a term from the following list. Terms may be used more than once.

10

11

ion

octet

12

13

anion

cation

1. An atom or group of atoms that has a positive or negative electric charge because it has lost or gained electrons is a(n) ion.
2. The sodium ion has 11 protons, 12 neutrons, and 10 electrons.
3. In most chemical reactions, atoms tend to match the outer electron configuration of the noble gases. This is called the octet rule.

Complete each statement below by writing the correct term or phrase.

4. An anion is an ion with a negative charge.
5. Valence electrons occupy the outermost energy level of an atom.
6. The chloride ion and the chlorine atom have different electron configurations.
7. A cation is an ion with a positive charge.
8. Many stable ions have an electron configuration of a noble gas.

Answer the following questions in the space provided.

9. How do the outer-shell electron configurations for ions of Group 1, Group 2, and Group 15, Group 16, and Group 17 elements compare with those of the noble gases?

They are the same. Group 1 + 2 ions have noble gas configuration of previous row. Group 15, 16 + 17 ions have noble gas configuration in their row.

Holt Chemistry Concept Review Answers Chap 1

Jessica J Manson



Holt Chemistry Concept Review Answers Chap 1:

Unveiling the Energy of Verbal Art: An Mental Sojourn through **Holt Chemistry Concept Review Answers Chap 1**

In some sort of inundated with monitors and the cacophony of quick transmission, the profound energy and mental resonance of verbal art usually fade in to obscurity, eclipsed by the regular barrage of sound and distractions. Yet, situated within the musical pages of **Holt Chemistry Concept Review Answers Chap 1**, a fascinating work of literary elegance that impulses with natural thoughts, lies an remarkable trip waiting to be embarked upon. Penned by way of a virtuoso wordsmith, this interesting opus courses readers on a mental odyssey, softly exposing the latent possible and profound affect embedded within the complex web of language. Within the heart-wrenching expanse of the evocative examination, we will embark upon an introspective exploration of the book is central styles, dissect their fascinating publishing model, and immerse ourselves in the indelible impact it leaves upon the depths of readers souls.

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