



basic education

Department:
Basic Education
REPUBLIC OF SOUTH AFRICA

PHYSICAL SCIENCES

GRADE 12

JUNE EXAMS- MEMO

2023

QUESTION 7 (start on a new page.)

- 7.1 Define an acid according to the Brønsted-Lowry theory.
- 7.2 The table below shows the ionisation constants (K_a) of two acids of equal concentrations.

NAME	FORMULA	K_a
Sulphuric acid	H_2SO_4	$1,3 \times 10^{-2}$
Sulphurous acid	H_2SO_3	$1,54 \times 10^{-2}$

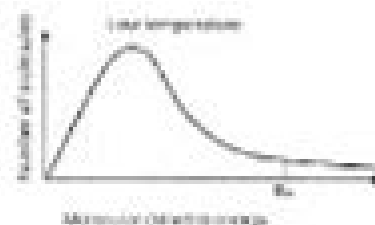
- 7.2.1 Which ONE of these acids will have a higher molar conductivity? Give a reason for this answer.
- 7.2.2 Write down a chemical reaction to show the ionisation of sulphuric acid in water.

A	B
C CH_3Cl	D Propanoic acid
E Propanol	F $\text{C}_2\text{H}_5\text{O}_2$

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- 5.3.4 Explain the shape of the graph and what happened after t_0 . (2)

- 5.4 The following Maxwell-Boltzmann distribution graph was obtained at a low temperature.



- 5.4.1 Copy the graph given above in your ANSWER BOOK. Use a dotted line and indicate on the graph how the distribution would change if a HIGHER TEMPERATURE.

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