

The screenshot shows a web-based application window titled 'Inventory'. The window has a red title bar and a navigation menu at the top with options like 'Home', 'Add', 'Edit', 'Delete', 'Print', and 'Filter'. Below the menu is a table with the following columns: 'Item No.', 'Description', and 'Status'. The table contains 15 rows of data, including items like 'PART & EQUIPMENT DEPOSIT', 'BELT DRIVE FOR JAW', 'AIR COMPRESSOR BELT', and 'CLEANER DIGITAL SYSTEM'. The 'Status' column contains values such as 'Open', 'In', 'Out', 'Total', and 'In-Stock'. The application is running in a browser, with the address bar showing 'http://localhost:8080/Inventory'. The bottom of the browser window shows a taskbar with a search bar and several icons.

Item No.	Description	Status
10000000000000000000	Part & equipment deposit, lat-dong	Open
BELT DRIVE FOR JAW	Compressor belt - 2P	In
BELT WITH AIRCOMP 5.0-10	Air compressor belt	In
BAND CONVEYOR 2000	Belt bearing 2000 - for conveyor roller	In
BAND CONVEYOR 2000	Belt bearing 2000	In
BAND EXTENSION 100	Belt bearing 100	In
BAND EXTENSION 400L	Belt bearing 400L	In
BAND WITH AIRCOMP 100	Belt bearing 100000 - for conveyor roller	In
BAND PULLEY EXTRA-CREW	Pulley bearing to rotate outer shell	In
BAND PULLEY EXTRA-CREW	Pulley bearing to rotate inner	In
CLEANER DIGITAL	Compressor digital system	Total
CLEANER DIGITAL TOP	Over Top Cleaner/Engineer	Open
CLEANER SHIP/SHOEN	Cleaner/Depositor	Open
CONCRETE TONER	Concrete stone	Open
CONDUIT COPPER 1/2"	Copper conduit	In
CONDUIT ULTRA	Ultra Conduit	In
CONCRETE	Conc. - Total 1. 814	Total

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web he concluded that at a constant pressure the volume of the gas would expand at a particular rate for each degree of temperature rise that being 1/267 per degree in 1874 victor regnault obtained better experimental results showing this number to be 1/273 and concluded that the pressure would approach zero at 1/273 degrees c

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web the volume of 1.00 mol of any gas at stp standard temperature 273.15 k and pressure 1 atm is measured to be 22.414 l we can substitute 101.325 kpa for pressure 22.414 l for volume and 273.15 k for temperature into the ideal gas equation and solve for r

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