

Analytical Chemistry MCQs: TYBSC SEM III

1] Gravimetric Analysis

1. If the solubility of calcium fluoride in water is S moles per litre, its solubility is given by

- (a) $2S$ (b) $2S^2$
(c) $4S^3$ (d) $4S^2$

2. Correct representation of solubility product of $\text{Ca}_3(\text{PO}_4)_2$ is

- (a) $[\text{Ca}^{2+}][\text{PO}_4^{3-}]^2$ (b) $[\text{Ca}^{2+}][\text{PO}_4^{3-}]$
(c) $[\text{Ca}][\text{PO}_4]$ (d) none of these

3. Dimethyl glyoxime is a reagent specific for the metal ion.

- (a) Al (b) Ni
(c) Fe (d) Cu

4. The solubility product expression for Hg_2Cl_2 is

- (a) $[\text{Hg}^{2+}][\text{Cl}^-]$ (b) $[\text{Hg}_2^{2+}][\text{Cl}^-]^2$
(c) $[\text{Hg}_2^{2+}][2\text{Cl}^-]$ (d) $[\text{Hg}^{+1}][\text{Cl}^-]$

5. A solution is called saturated if

- (a) ionic product < solubility product.
(b) ionic product > solubility product.
(c) ionic product = solubility product.
(d) none of these

6. Solubility of alcohol in water is due to

- (a) ionic bonding
(b) covalent bonding
(c) Metallic bonding
(d) Hydrogen bonding

7. Precipitation takes place when the product of concentration of ions

- (a) Exceeds their solubility product
(b) Is less than their solubility product
(c) Equal to their solubility product
(d) None of these

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KJ Lindholm-Leary



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