NAMING BINARY COMPOUNDS (IONIC)

Name _____

Name the following ionic compounds using Roman numerals where necessary.

1. BaCl, _______ 11. K,S

2. NaF ______ 12. CrCl,

______ 13. CrCl, 3. Ag,O

14. CaO 4. CuBr

_____ 15. Ba,P, 5. CuBr.

6. FeO

______ 16. Hg₂l₂

7. Fe,O, ______ 17. Na,O

8. MgS ______ 18. BeS

9. Al,O,

______ 19. MnO

10. Cal.

______ 20. Mn₂O₃ ______

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United States. National Bureau of Standards, Betty L. Oberholtzer

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Prentice Hall Chemistry ,2000 **Glencoe Science** McGraw-Hill Staff,2001-06 **Chemistry** John S. Phillips, Cheryl Wistrom.2000 Science Software Quarterly ,1985 Master Analytical Manual: Ionic methods Oak Ridge National Laboratory. Analytical Chemistry Division, 1959 Chemistry, Student Study Guide James E. Brady, Fred Senese, 2008-01-28 The image on the front cover depicts a carbon nanotube emerging from a glowing plasma of hydrogen and carbon as it forms around particles of a metal catalyst Carbon nanotubes are a recently discovered allotrope of carbon Three other allotropes of carbon buckyballs graphite and diamond are illustrated at the left as is the molecule methane CH4 from which nanotubes and buckyballs can be made The element carbon forms an amazing number of compounds with structures that follow from simple methane found in natural gas to the complex macromolecules that serve as the basis of life on our planet The study of chemistry also follows from the simple to the more complex and the strength of this text is that it enables students with varied backgrounds to proceed together to significant levels of achievement The Success Manual for General Chemistry Elizabeth Kean, Catherine Middlecamp, 1986 Chemical Principles Peter Atkins, Loretta

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