

Name: _____ Date: _____ Period: _____

NOVA: Hunting the Elements

Part 1: Basic Chemistry

Periodic Table of Elements

Use the periodic table below to help you answer some questions covered in the documentary.

| | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | |
|--------------------------|--|---------------------|--|-----------------------|--|---------------------|--|----------------------|--|-----------------------|--|-----------------------|--|---------------------|--|------------------------|--|----------------------|--|----------------------|--|------------------------|--|----------------------|--|----------------------|--|---------------------|--|----------------------|--|--------------------|--|--------------------|--|-------------------|--|
| 1 H 1.00794 | | | | | | | | | | | | | | | | | | 2 He 4.002602 | | | | | | | | | | | | | | | | | | | |
| 3 Li 6.941 | | 4 Be 9.012182 | | | | | | | | | | | | | | 5 B 10.811 | | 6 C 12.011 | | 7 N 14.006434 | | 8 O 15.999 | | 9 F 18.9984032 | | 10 Ne 20.17976 | | | | | | | | | | | |
| 11 Na 22.98976928 | | 12 Mg 24.304 | | | | | | | | | | | | | | 13 Al 26.9815386 | | 14 Si 28.08558 | | 15 P 30.973762 | | 16 S 32.06 | | 17 Cl 35.453 | | 18 Ar 39.948 | | | | | | | | | | | |
| 19 K 39.0983 | | 20 Ca 40.078 | | 21 Sc 44.955912 | | 22 Ti 47.88 | | 23 V 50.9415 | | 24 Cr 51.9961 | | 25 Mn 54.938045 | | 26 Fe 55.845 | | 27 Co 58.933195 | | 28 Ni 58.6934 | | 29 Cu 63.546 | | 30 Zn 65.38 | | 31 Ga 69.723 | | 32 Ge 72.63 | | 33 As 74.9216 | | 34 Se 78.96 | | 35 Br 79.904 | | 36 Kr 83.80 | | | |
| 37 Rb 85.4678 | | 38 Sr 87.62 | | 39 Y 88.90584 | | 40 Zr 91.224 | | 41 Nb 92.90638 | | 42 Mo 95.94 | | 43 Tc 98 | | 44 Ru 101.07 | | 45 Rh 102.9055 | | 46 Pd 106.42 | | 47 Ag 107.8682 | | 48 Cd 112.411 | | 49 In 114.818 | | 50 Sn 118.710 | | 51 Sb 121.757 | | 52 Te 127.6 | | 53 I 126.905 | | 54 Xe 131.29 | | | |
| 55 Cs 132.90545196 | | 56 Ba 137.327 | | 57-70 * | | 71 Lu 174.967 | | 72 Hf 178.49 | | 73 Ta 180.94788 | | 74 W 183.84 | | 75 Re 186.207 | | 76 Os 190.23 | | 77 Ir 192.222 | | 78 Pt 195.084 | | 79 Au 196.966569 | | 80 Hg 200.59 | | 81 Tl 204.3833 | | 82 Pb 207.2 | | 83 Bi 208.9804 | | 84 Po [209] | | 85 At [210] | | 86 Rn [222] | |
| 87 Fr [223] | | 88 Ra [226] | | 89-102 *-* | | 103 Lr [260] | | 104 Rf [261] | | 105 Db [262] | | 106 Sg [266] | | 107 Bh [264] | | 108 Hs [277] | | 109 Mt [268] | | 110 Uun [289] | | 111 Uuu [288] | | 112 Uub [285] | | 113 Uut [284] | | 114 Uuq [289] | | | | | | | | | |

* Lanthanide series

| | | | | | | | | | | | | | |
|-----------------------|--------------------|-----------------------|---------------------|-------------------|--------------------|---------------------|--------------------|-----------------------|----------------------|-----------------------|---------------------|-----------------------|-----------------------|
| 57 La 138.90547 | 58 Ce 140.12 | 59 Pr 140.90766 | 60 Nd 144.242 | 61 Pm [145] | 62 Sm 150.36 | 63 Eu 151.964 | 64 Gd 157.25 | 65 Tb 158.92532 | 66 Dy 162.5001 | 67 Ho 164.93033 | 68 Er 167.259 | 69 Tm 168.93032 | 70 Yb 173.05468 |
|-----------------------|--------------------|-----------------------|---------------------|-------------------|--------------------|---------------------|--------------------|-----------------------|----------------------|-----------------------|---------------------|-----------------------|-----------------------|

** Actinide series

| | | | | | | | | | | | | | |
|-------------------|----------------------|-----------------------|----------------------|-------------------|-------------------|-------------------|-------------------|-------------------|-------------------|-------------------|--------------------|--------------------|--------------------|
| 89 Ac [227] | 90 Th 232.0377 | 91 Pa 231.03688 | 92 U 238.02891 | 93 Np [237] | 94 Pu [244] | 95 Am [243] | 96 Cm [247] | 97 Bk [247] | 98 Cf [251] | 99 Es [252] | 100 Fm [257] | 101 Md [258] | 102 No [259] |
|-------------------|----------------------|-----------------------|----------------------|-------------------|-------------------|-------------------|-------------------|-------------------|-------------------|-------------------|--------------------|--------------------|--------------------|

Gold - Au

- Write the number of subatomic particles in gold:
 - Protons: **79**
 - Neutrons: **118**
 - Electrons: **79**
- Give one property of gold.
shiny, malleable (soft), high density, electrically conductive
- How much gold is in one ton of the mined rock?
1 ounce

Huntinh The Elements Packet Answers

G Thomas



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