Chemistry Form 5 Chapter 1

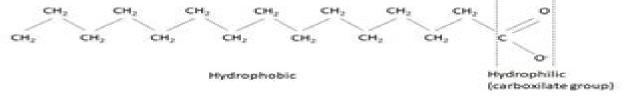
Rate of Reaction

- · Rate of reaction is the change in quantity of reactant or product against time
- Method used to measure the volume of hydrogen gas is water displacement method

Fast reaction	Slow reaction
Short time taken for the reaction	Long time taken for the reaction
High rate of reaction	Low rate of reaction
Sodium + Water	Copper(II) oxide + Hydrogen gas
Neutralisation	Fermentation
Percipitation	Rusting

Effective reaction

- Effective reaction are collisions between the particles of reactants that cause a reaction
- Occur when the particles of reactants have energy that is equal or higher than activation energy.
- Effective reaction increases, the rate of reaction increases.



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