

Form One Chemistry Exam

Duration: 1 hour 30 minutes

Instructions:

- Answer all questions.
- Show all working where applicable.
- Write your answers clearly.

SECTION A: Multiple Choice Questions (1 mark each)

- 1. What is the smallest unit of matter?
A. Atom B. Molecule C. Compound D. Element
- 2. Which one of the following is a compound?
A. Iron B. Oxygen C. Water D. Copper
- 3. Which state of matter has a definite volume but no fixed shape?
A. Solid B. Liquid C. Gas D. Plasma
- 4. The chemical symbol for Sodium is:
A. S B. N C. Na D. So
- 5. Which of the following is NOT a physical change?
A. Melting of ice B. Boiling water C. Rusting of iron D. Dissolving salt in water

SECTION B: Short Answer Questions (2 marks each)

6. Define chemistry.
7. List two differences between a physical change and a chemical change.
8. Name the three common states of matter and give one example for each.
9. What is a mixture? Give two examples.
10. What is the main difference between an element and a compound.

SECTION C: Structured Questions

11. (a) State any three laboratory safety rules. (3 marks)
(b) Why is it dangerous to taste chemicals in the lab? (2 marks)
12. (a) Name any four apparatus used in the chemistry laboratory. (2 marks)
(b) State one use for each apparatus mentioned above. (2 marks)
13. The diagram below shows the separation of a mixture of sand and salt:
(a) What method is used to separate the mixture? (1 mark)
(b) List the steps involved in the separation. (4 marks)

Form One Chemistry Revision Questions

Shikha Goel, Saleha Parvez



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Higher Chemistry: Revision Notes and Questions John Anderson, 2017-03-06 Exam Board SQA Level Higher Subject Chemistry First Teaching September 2014 First Exam Summer 2015 The perfect revision companion to the Higher Chemistry textbook this newly written resource reflects the very latest SQA curriculum This book offers bite sized facts worked examples and a variety of question types to support and reinforce learning and practice for the final examination Dedicated sections on numeracy and researching chemistry Provides a broad range of questions including open ended and past SQA exam questions Answers section allows students to check their understanding of the standards required for Higher level success

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- Chapter 5: Conclusion

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