

THERMOCHEMISTRY

SAMPLE TEST

Multiple choice:

- 10) A certain amount of substance reacts exothermically with 2.700 g of water.
 - a) 284.400 J
 - b) 284.400 J
 - c) -284.400 J
 - d) -284.400 J
- 11) A heat transfer of 1000 J occurs and heat and volume changes are negligible. If 10 g of water (from 20.000 °C to 20.000 J/g °C) is heated.
 - a) 2000 J/mol
 - b) 2000 J/mol
 - c) -2000 J/mol
 - d) -2000 J/mol
- 12) A heat of fusion of 6.000 J/mol is needed to melt 1.000 g of ice. If the heat is supplied by raising the temperature of 1000 g of water (from 20.000 °C to 20.000 J/g °C).
 - a) 20000 J/g °C
 - b) 20000 J/g °C
 - c) -20000 J/g °C
 - d) -20000 J/g °C
- 13) A substance requires heat of 10.000 J/g °C. If 1000 g of substance is heated 1.000 °C, what amount of heat is transferred?
 - a) 1.000 °C
 - b) 1.000 °C
 - c) 1.000 °C
 - d) 1.000 °C
- 14) A certain amount of substance reacts exothermically with 2.700 g of water. If the heat is supplied by raising the temperature of 1000 g of water (from 20.000 °C to 20.000 J/g °C).
 - a) 20000 J/g °C
 - b) 20000 J/g °C
 - c) -20000 J/g °C
 - d) -20000 J/g °C

Multiple Choice Thermochemistry Questions

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