

OBSERVATION MANUAL

FOR

**MICROPROCESSORS AND
MICROCONTROLLERS LAB
FOR B.TECH III YEAR I SEMESTER**

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ROLL NO.:

SECTION:

ACADEMIC YEAR: 2018 - 2019



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The laboratory portion of a chemistry class can be a concern for teachers with limited lab facilities This includes teachers in private schools public schools charter schools and home schools This manual and the accompanying kit are an effort to help solve this problem The laboratory exercises have been designed with three goals in mind 1 educational challenge 2 safety and 3 convenience for the teacher The kits intended for the laboratory portion of the course are based on the microscale method This approach to chemistry gives students a lab experience as good as or better than the traditional methods but uses about 1 100th of the chemicals The experiments are much safer and disposal much easier The chemical solutions are pre mixed and in dropping bottles that give constant drop size This eliminates the need to mix solutions greatly reduces spills and reduces set up time to a few minutes Introduction Lab Melting Points Super Cooling 1 Empirical Formula 2 Analysis of Hydrates 3 Molar Mass by Titration 4 Freezing Point Depression 5 Gas Laws Boyle s Law 6 Gas Laws Charles s Law 7 Molar Volume of a Gas 8 A Standard Acid and a Standardized Base 9 A Microscale Titration 10 A Weak Acid Strong Base Titration 11 Oxidation Reduction 12 Mole Ratios 13 Double Replacement Reactions 14 Solubility Product Constant 15 pH and pH Indicators 16 Reaction Rates The Effect of Concentration 17 Reaction Rates The Effects of Temperature and Particle Size 18 Radioactive Decay 19 Enthalpy of Fusion of Ice 20 Decomposition of H₂O and NaCl 21 Properties of Cations and Anions 22 Synthesis of a Coordination Compound 23 Synthesis and Analysis of Aspirin 24 Gravimetric Analysis 25 Colorimetry 26 Paper Chromatography 27 A Buffer Solution 28 Electrical Conductivity of Several Solutions 29 Electrochemistry Galvanic Cells

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