

(b) Ethanoic acid reacts with another alcohol, **Y**, to produce an ester **Z**.

(i) Alcohol **Y** has molar mass 74 g mol^{-1} and the following composition by mass:

carbon, C = 64.9%

hydrogen, H = 13.5%

oxygen, O = 21.6%.

Use all these data to confirm that the molecular formula for **Y** is $\text{C}_4\text{H}_{10}\text{O}$.
Show your working.

(2)

(ii) Draw the displayed formulae of the **four** possible structures of alcohol **Y**.

(2)

Alcohol 1	Alcohol 2
Alcohol 3	Alcohol 4



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 - The Rise of Digital Reading January 2015 Ial Edexcel Maths Paper
 - Advantages of eBooks Over Traditional Books
2. Identifying January 2015 Ial Edexcel Maths Paper
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an January 2015 Ial Edexcel Maths Paper
 - User-Friendly Interface
4. Exploring eBook Recommendations from January 2015 Ial Edexcel Maths Paper
 - Personalized Recommendations
 - January 2015 Ial Edexcel Maths Paper User Reviews and Ratings
 - January 2015 Ial Edexcel Maths Paper and Bestseller Lists

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 - January 2015 Ial Edexcel Maths Paper Budget-Friendly Options
6. Navigating January 2015 Ial Edexcel Maths Paper eBook Formats
 - ePub, PDF, MOBI, and More
 - January 2015 Ial Edexcel Maths Paper Compatibility with Devices
 - January 2015 Ial Edexcel Maths Paper Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of January 2015 Ial Edexcel Maths Paper
 - Highlighting and Note-Taking January 2015 Ial Edexcel Maths Paper
 - Interactive Elements January 2015 Ial Edexcel Maths Paper
8. Staying Engaged with January 2015 Ial Edexcel Maths Paper
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers January 2015 Ial Edexcel Maths Paper
9. Balancing eBooks and Physical Books January 2015 Ial Edexcel Maths Paper
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection January 2015 Ial Edexcel Maths Paper
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine January 2015 Ial Edexcel Maths Paper
 - Setting Reading Goals January 2015 Ial Edexcel Maths Paper
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of January 2015 Ial Edexcel Maths Paper
 - Fact-Checking eBook Content of January 2015 Ial Edexcel Maths Paper
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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