

- Directions:**
 1. An element may be assigned a positive or negative charge depending on whether it is a metal or nonmetal.
 2. An atom with a positive charge is a cation.
 3. An atom with a negative charge is an anion.
 4. The number of protons in an atom is the atomic number.

Ion Practice Set

1. Write the symbol for the ion that has the same number of protons as the element but a different number of electrons.
 2. Write the symbol for the ion that has the same number of protons as the element but a different number of electrons.

Complete the following table, using the periodic table in the back of your book.

	ELEMENT NAME	ION SYMBOL	NUMBER OF PROTONS	NUMBER OF ELECTRONS	CHARGE OF ION (POSITIVE OR NEGATIVE)	TYPE OF ION
1	Sodium	Na^+	11	10	positive	cation
2	Chlorine	Cl^-	17	18	negative	anion
3	Strontium	Sr^{2+}	38	36	positive	cation
4	Fluorine	F^-	9	10	negative	anion
5	Calcium	Ca^{2+}	20	18	positive	cation
6	Oxygen	O^{2-}	8	10	negative	anion
7	Aluminum	Al^{3+}	13	10	positive	cation
8	Hydrogen	H^+	1	0	positive	cation
9	Lithium	Li^+	3	2	positive	cation
10	Magnesium	Mg^{2+}	12	10	positive	cation
11	Sulfur	S^{2-}	16	18	negative	anion
12	Selenium	Se^{2-}	34	36	negative	anion
13	Hydrogen	H^+	1	0	positive	cation
14	Barium	Ba^{2+}	56	54	positive	cation
15	Strontium	Sr^{2+}	38	36	positive	cation
16	Calcium	Ca^{2+}	20	18	positive	cation

Ions And Isotope Practice Set Answer Key

Clemens Wendtner



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Table of Contents Ions And Isotope Practice Set Answer Key

1. Understanding the eBook Ions And Isotope Practice Set Answer Key
 - The Rise of Digital Reading Ions And Isotope Practice Set Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Ions And Isotope Practice Set Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Ions And Isotope Practice Set Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Ions And Isotope Practice Set Answer Key

- Personalized Recommendations
- Ions And Isotope Practice Set Answer Key User Reviews and Ratings
- Ions And Isotope Practice Set Answer Key and Bestseller Lists
- 5. Accessing Ions And Isotope Practice Set Answer Key Free and Paid eBooks
 - Ions And Isotope Practice Set Answer Key Public Domain eBooks
 - Ions And Isotope Practice Set Answer Key eBook Subscription Services
 - Ions And Isotope Practice Set Answer Key Budget-Friendly Options
- 6. Navigating Ions And Isotope Practice Set Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Ions And Isotope Practice Set Answer Key Compatibility with Devices
 - Ions And Isotope Practice Set Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Ions And Isotope Practice Set Answer Key
 - Highlighting and Note-Taking Ions And Isotope Practice Set Answer Key
 - Interactive Elements Ions And Isotope Practice Set Answer Key
- 8. Staying Engaged with Ions And Isotope Practice Set Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Ions And Isotope Practice Set Answer Key
- 9. Balancing eBooks and Physical Books Ions And Isotope Practice Set Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Ions And Isotope Practice Set Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Ions And Isotope Practice Set Answer Key
 - Setting Reading Goals Ions And Isotope Practice Set Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Ions And Isotope Practice Set Answer Key

- Fact-Checking eBook Content of Ions And Isotope Practice Set Answer Key
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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