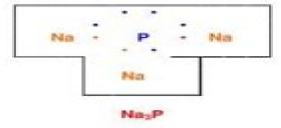
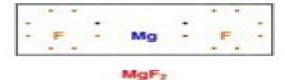
Using Lewis dot diagrams, demonstrate how each of the following ionic compounds are formed. The first step is to find out how many valence electrons each element has, then draw the Lewis dot diagram. Use the confetti pieces to help with each diagram until it becomes easier to do. Write the chemical symbol for the ionic compound formed.

Sodium and phosphorus



2. Magnesium and fluorine

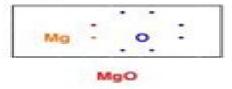


3. Potassium and bromine



KBr

Magnesium and oxygen



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