

Radioactive Isotopes

1. Calculate the half-life of a radioactive isotope if 100 g decays to 50 g in 1 hour.
2. A sample of a radioactive isotope has an initial activity of 80,000 becquerels. If the half-life of the isotope is 10 days, calculate its activity after 30 days.
3. A sample of a radioactive isotope contains 100 g of radioactive material. Calculate its mass after 10 half-lives.
4. A sample of a radioactive isotope has an activity of 10,000 becquerels. If the half-life of the isotope is 2 days, calculate its activity after 5 days.
5. If a sample of a radioactive isotope contains 1,000 g of radioactive material, how many grams will remain after 5 half-lives?
6. A sample of a radioactive isotope has an activity of 20,000 becquerels. If the half-life of the isotope is 5 days, calculate its activity after 10 days.
7. If a sample of a radioactive isotope contains 500 g of radioactive material, how many grams will remain after 3 half-lives?
8. A sample of a radioactive isotope has an activity of 60,000 becquerels. If the half-life of the isotope is 8 days, calculate its activity after 16 days.
9. A sample of a radioactive isotope has an initial activity of 100,000 becquerels. If the half-life of the isotope is 7 days, calculate its activity after 14 days.
10. If a sample of a radioactive isotope contains 200 g of radioactive material, how many grams will remain after 4 half-lives?

Half Life Of Radioactive Isotopes Worksheet Page 35

Answers

JA Banks



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Half Life Of Radioactive Isotopes Worksheet Page 35 Answers Book Review: Unveiling the Power of Words

In some sort of driven by information and connectivity, the power of words has be evident than ever. They have the ability to inspire, provoke, and ignite change. Such is the essence of the book **Half Life Of Radioactive Isotopes Worksheet Page 35 Answers**, a literary masterpiece that delves deep into the significance of words and their effect on our lives. Compiled by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we will explore the book is key themes, examine its writing style, and analyze its overall effect on readers.

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