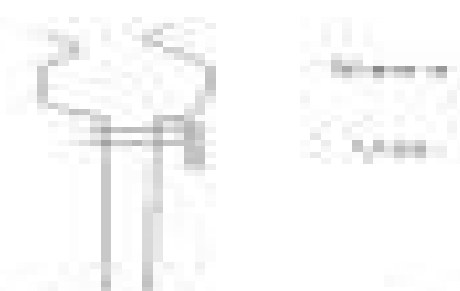


Figure 1: Chemistry: Electrode Potentials and Diagrams

Question 1:

(a) The following are standard conventional half-cells. Label them accordingly (20%)



(b) What are measures to be taken to ensure accuracy and results of pH?

Answer 1:

There is a standard solution of H+ ions, usually, hydrochloric acid 0.1 mol dm⁻³.

(a) Standard electrode and standard of hydrogen gas would also be used in the standard solution.

(b) There is a reason for the choice of the standard.

(c) Standard electrode would use the standard hydrogen gas which is standard for the hydrogen.

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DP Hallahan



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