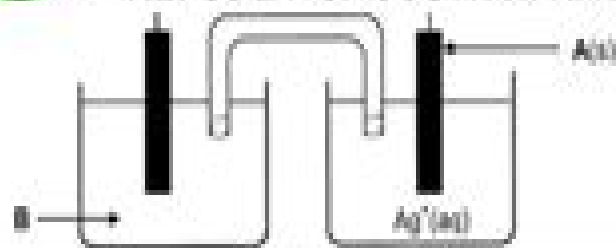




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Consider the equations for reactions I to III below.

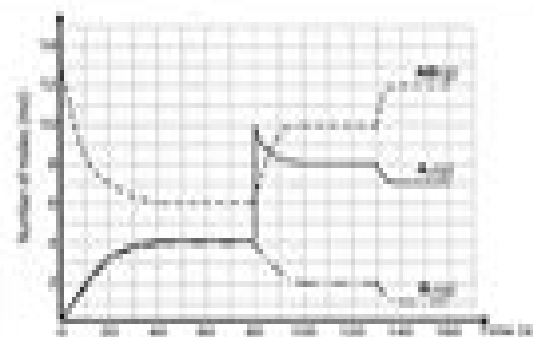
A and B represent organic compounds that are POSITIONAL ISOMERS.
X is an inorganic product.

I	$\text{CH}_3\text{CH}_2\text{CH}(\text{OH})\text{CH}_3 + \text{HCl} \rightarrow \text{A} + \text{B}$
II	$\text{A} \xrightarrow[\Delta]{\text{H}_2\text{O}} \text{CH}_3\text{CH}_2\text{CH}_2\text{CH}(\text{OH})\text{CH}_3 + \text{X}$
III	$\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}(\text{OH})\text{CH}_3 \rightarrow \text{CH}_3\text{CH}_2\text{CH}(\text{OH})\text{CH}_3 + \text{H}_2\text{O}$



He later wrote summarising the reaction conditions and results of the experiment

Run	VOL. H ₂ O NaOH(aq) (cm ³)	VOL. H ₂ O HCl(aq) (cm ³)	CONCENTRATION NaOH(aq) (mol dm ⁻³)	Time (s)
1	10	0	0.11	25.4
2	10	10	0.10	26.7
3	10	20	#	33.5



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