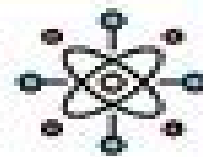


CHEMICAL BONDS



IONIC, COVALENT, & METALLIC

FORMS OF CHEMICAL BONDS
Bonding: combination of atoms

Atoms combine to form molecules or compounds by sharing or transferring electrons. This process is called bonding. The resulting molecule or compound has properties different from the individual atoms.

Types of Bonds:

- Covalent Bond:** Formed by sharing electrons between two atoms. Example: H_2 , H_2O , CO_2 .
- Ionic Bond:** Formed by the transfer of electrons from one atom to another, resulting in oppositely charged ions that attract each other. Example: $NaCl$, MgO .
- Metallic Bond:** Formed by the attraction between metal cations and a sea of delocalized electrons. Example: Fe , Cu , Al .

Properties of Bonds:

- Covalent bonds are strong and directional.
- Ionic bonds are strong and non-directional.
- Metallic bonds are strong and non-directional.

Science Real

A simplified periodic table of elements, color-coded by groups.

IONIC BONDING

Occurs between a metal and a non-metal. The metal atom loses one or more electrons to become a positively charged cation, and the non-metal atom gains those electrons to become a negatively charged anion. The resulting compound is held together by the electrostatic attraction between the oppositely charged ions.

COVALENT BONDING

Occurs between two non-metal atoms. They share one or more pairs of electrons to form a stable molecule. The shared electrons are represented by lines (single, double, or triple bonds) or dots (electron dot diagrams).

Diagram illustrating ionic bonding: A red circle with a minus sign ($-$) and a yellow circle with a plus sign ($+$) are shown. Dashed lines connect them, representing the electrostatic attraction between the ions.

Diagram illustrating covalent bonding: Two atoms are shown with overlapping electron shells. The overlapping region contains two dots, representing a shared pair of electrons. The text on the diagram states: "If each atom already has an electron with the other, both atoms have eight valence electrons."

Diagram illustrating metallic bonding: A cluster of metal atoms is shown, each with a nucleus and a sea of delocalized electrons surrounding them. The text on the diagram states: "Metal atoms are held together by the attraction between the positive metal ions and the sea of delocalized electrons."

GUIDED READING ACTIVITY

Ionic And Metallic Bonding Guided Reading

EW Minium

A decorative graphic element consisting of a light blue horizontal bar with a rounded right end, and a red circular shape partially visible behind it.

Ionic And Metallic Bonding Guided Reading:

Unveiling the Energy of Verbal Beauty: An Psychological Sojourn through **Ionic And Metallic Bonding Guided Reading**

In a global inundated with monitors and the cacophony of quick transmission, the profound power and mental resonance of verbal artistry frequently fade into obscurity, eclipsed by the regular assault of noise and distractions. However, located within the musical pages of **Ionic And Metallic Bonding Guided Reading**, a captivating function of fictional brilliance that impulses with fresh emotions, lies an memorable journey waiting to be embarked upon. Written with a virtuoso wordsmith, this interesting opus instructions visitors on an emotional odyssey, gently exposing the latent potential and profound influence stuck within the complex internet of language. Within the heart-wrenching expanse of this evocative examination, we can embark upon an introspective exploration of the book is main styles, dissect its charming publishing model, and immerse ourselves in the indelible effect it leaves upon the depths of readers souls.

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