



IONIC AND METALLIC BONDING

Chapter Test A

A. Matching

Match each definition with the correct term. Write the letter of the correct term in the space provided.

Column A

- _____ 1. an electron lost or gained
- _____ 2. an ion's charge
- _____ 3. value of electric force
- _____ 4. positive ion
- _____ 5. electrovalency
- _____ 6. anion (neg)
- _____ 7. cation
- _____ 8. compound
- _____ 9. formula unit
- _____ 10. coordination number
- _____ 11. crystal lattice

Column B

- _____ 12. ion with same number of electrons as neutral atom (a noble gas)
- _____ 13. net positive charge of positive ions exceeding that of negative ions
- _____ 14. the more electronegative element changed into negative
- _____ 15. the smallest combination of positive and negative ions
- _____ 16. a diagram of valence electrons around the symbol of an element
- _____ 17. compound of positive and negative ions
- _____ 18. the most electronegative
- _____ 19. all elements in a group in a period along a horizontal row
- _____ 20. the number representing total of electrons during combination of 2 atoms
- _____ 21. atoms are held together by forces of attraction (represented as a lattice)
- _____ 22. fixed definite number ratio of positive and negative ions

B. Multiple Choice

Write the letter of the correct answer in the space provided.

- _____ 23. Which group contains elements that do not form ionic compounds? A. Group 1
B. Group 2
C. Group 17
D. Group 18

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