

Intro to Chemical Reactions:

Test Review Answer Key



1. What do the symbols (s), (l), (g), and (aq) represent? (s) = solid, (l) = liquid, (g) = gas, and (aq) = aqueous
2. Why must chemical equations be balanced? Chemical equations must be balanced to satisfy the Law of Conservation of Mass. The number of atoms of the reactant side must equal the number of atoms on the product side of a chemical equation.
3. How can we be certain that a chemical reaction has taken place? Color change, temperature change, formation of a precipitate, production of gas (bubbles), and light emission.
4. Balance the following chemical equations:
 - a. $1 \text{ AsCl}_3 + 3 \text{ H}_2\text{O} \rightarrow 3 \text{ HCl} + 1 \text{ As(OH)}_3$
 - b. $4 \text{ Na} + 1 \text{ O}_2 \rightarrow 2 \text{ Na}_2\text{O}$
 - c. $1 \text{ Cu} + 2 \text{ H}_2\text{SO}_4 \rightarrow 1 \text{ CuSO}_4 + 1 \text{ SO}_2 + 2 \text{ H}_2\text{O}$
 - d. $2 \text{ Sb} + 3 \text{ H}_2\text{O} \rightarrow 1 \text{ Sb}_2\text{O}_3 + 3 \text{ H}_2$
 - e. $1 \text{ La(NO}_3)_3 + 3 \text{ NaOH} \rightarrow 1 \text{ La(OH)}_3 + 3 \text{ NaNO}_3$
 - f. $1 \text{ Ca(AlO}_2)_2 + 8 \text{ HCl} \rightarrow 2 \text{ AlCl}_3 + 1 \text{ CaCl}_2 + 4 \text{ H}_2\text{O}$
 - g. $1 \text{ C}_4\text{H}_8 + 6 \text{ O}_2 \rightarrow 4 \text{ CO}_2 + 4 \text{ H}_2\text{O}$
 - h. $2 \text{ C}_3\text{H}_4\text{O}_3 + 7 \text{ O}_2 \rightarrow 6 \text{ CO}_2 + 8 \text{ H}_2\text{O}$
5. Identify the following types of chemical reactions:
 - a. $2 \text{ H}_2 + \text{O}_2 \rightarrow 2 \text{ H}_2\text{O}$ **Synthesis or Combustion**
 - b. $2 \text{ H}_2\text{O} \rightarrow 2 \text{ H}_2 + \text{O}_2$ **Decomposition**
 - c. $\text{Zn} + \text{H}_2\text{SO}_4 \rightarrow \text{ZnSO}_4 + \text{H}_2$ **Single Replacement**
 - d. $2 \text{ CO} + \text{O}_2 \rightarrow 2 \text{ CO}_2$ **Synthesis or Combustion**
 - e. $2 \text{ HgO} \rightarrow 2 \text{ Hg} + \text{O}_2$ **Decomposition**
 - f. $2 \text{ KBr} + \text{Cl}_2 \rightarrow 2 \text{ KCl} + \text{Br}_2$ **Single Replacement**
 - g. $\text{CaO} + \text{H}_2\text{O} \rightarrow \text{Ca(OH)}_2$ **Synthesis**
 - h. $\text{AgNO}_3 + \text{NaCl} \rightarrow \text{AgCl} + \text{NaNO}_3$ **Double Replacement**
 - i. $2 \text{ H}_2\text{O}_2 \rightarrow 2 \text{ H}_2\text{O} + \text{O}_2$ **Decomposition**
 - j. $\text{Ca(OH)}_2 + \text{H}_2\text{SO}_4 \rightarrow \text{CaSO}_4 + 2 \text{ H}_2\text{O}$ **Double Replacement**

Intro To Chemical Reactions Test

Terry C. Jones



Intro To Chemical Reactions Test:

Intro To Chemical Reactions Test: Bestsellers in 2023 The year 2023 has witnessed a remarkable surge in literary brilliance, with numerous captivating novels captivating the hearts of readers worldwide. Lets delve into the realm of top-selling books, exploring the engaging narratives that have captivated audiences this year. Intro To Chemical Reactions Test : Colleen Hoover's "It Ends with Us" This heartfelt tale of love, loss, and resilience has captivated readers with its raw and emotional exploration of domestic abuse. Hoover skillfully weaves a story of hope and healing, reminding us that even in the darkest of times, the human spirit can prevail. Uncover the Best : Taylor Jenkins Reids "The Seven Husbands of Evelyn Hugo" This captivating historical fiction novel unravels the life of Evelyn Hugo, a Hollywood icon who defies expectations and societal norms to pursue her dreams. Reids captivating storytelling and compelling characters transport readers to a bygone era, immersing them in a world of glamour, ambition, and self-discovery. Intro To Chemical Reactions Test : Delia Owens "Where the Crawdads Sing" This mesmerizing coming-of-age story follows Kya Clark, a young woman who grows up alone in the marshes of North Carolina. Owens weaves a tale of resilience, survival, and the transformative power of nature, entrancing readers with its evocative prose and mesmerizing setting. These top-selling novels represent just a fraction of the literary treasures that have emerged in 2023. Whether you seek tales of romance, adventure, or personal growth, the world of literature offers an abundance of captivating stories waiting to be discovered. The novel begins with Richard Papen, a bright but troubled young man, arriving at Hampden College. Richard is immediately drawn to the group of students who call themselves the Classics Club. The club is led by Henry Winter, a brilliant and charismatic young man. Henry is obsessed with Greek mythology and philosophy, and he quickly draws Richard into his world. The other members of the Classics Club are equally as fascinating. Bunny Corcoran is a wealthy and spoiled young man who is always looking for a good time. Charles Tavis is a quiet and reserved young man who is deeply in love with Henry. Camilla Macaulay is a beautiful and intelligent young woman who is drawn to the power and danger of the Classics Club. The students are all deeply in love with Morrow, and they are willing to do anything to please him. Morrow is a complex and mysterious figure, and he seems to be manipulating the students for his own purposes. As the students become more involved with Morrow, they begin to commit increasingly dangerous acts. The Secret History is a brilliant and gripping novel that will keep you speculating until the very end. The novel is a cautionary tale about the dangers of obsession and the power of evil.

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