SHORT FORMULA (GYAN SUTRA)

CHEMISTRY

PHYSICAL CHEMISTRY

ATOMIC STRUCTURE

Estimation of closest distance of approach (derivation) of α -particle: $R = \frac{4KZe^2}{m_eV_e^2}$

The radius of a nucleus : $R = R_a (A)^{1/2} cm$

Planck's Quantum Theory: Energy of one photon = $hv = \frac{hc}{\lambda}$

Photoelectric Effect: $h_V = h_{V_0} + \frac{1}{2} m_e v^2$

Bohr's Model for Hydrogen like atoms:

1. $m_{VT} = n \frac{h}{2\pi}$ (Quantization of angular momentum)

2.
$$E_n = -\frac{E_1}{n^2} z^2 = 2.176 \times 10^{-18} \frac{z^2}{n^2}$$
 J/stom = 13.6 $\frac{z^2}{n^2}$ eV ; $E_1 = \frac{-2x^2 \text{ me}^4}{n^2}$

3.
$$r_n = \frac{n^2}{Z} \times \frac{h^2}{4\pi^2 e^2 m} = \frac{0.529 \times n^2}{Z} A$$
 4. $v = \frac{2\pi z e^2}{nh} = \frac{2.18 \times 10^8 \times z}{n} \text{ m/s}$

De-Broglie wavelength:

$$\lambda = \frac{h}{mc} = \frac{h}{p}$$
 (for photon)

Wavelength of emitted photon:

$$\frac{1}{\lambda} = \varphi = RZ^2 \left[\frac{1}{n_1^2} - \frac{1}{n_2^2} \right]$$

No. of photons emitted by a sample of H atom:

<u>**Iit Jee Physics Chemistry Maths Formula Booklet**</u>

Helmut J. Fischbeck, Kurt Fischbeck

Iit Jee Physics Chemistry Maths Formula Booklet:

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In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is really astonishing. Within the pages of "**Iit Jee Physics Chemistry Maths Formula Booklet**," an enthralling opus penned by a very acclaimed wordsmith, readers set about an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve into the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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