Organic Chemistry

PETROLEUM : FOSSIL FUEL

Petroloum - dead organisms

that but to ocean their surfed

under thick satisfies

tisgle protiure closely concreted

them to petroloum

Fostil fuel are permitturn, teal and natural gas

Notweel gas - methode often found with petroloum High temperature and pressure round the compounds to break down to gas Cost - remoins of tush
respectation buried under thick
sestiment
Pressure and heat stooly
converted into cost

moment is petrolessen?

- · A smelly mixture of hundreds of different compounds
- · Organic compound mostly nydrocarbons
- . A non-renewable resource
- Refine petroleum , corried out by fractional distillation to separate compounds into groups with molecules of similar size

carbon atoms band to each other hydrogen band to carbon atoms

H H H H H H
H -C - C - C - C - H
H H H H

Fractions

Boiling point	Flows	Volettle	Easily burn?	Stre of molecules
wp to 100-C	very runny	volotile	very easy	small
100" - 150"	nummy	less volutile	easy	
150°C - 200°C	not very runny	even less volatile	not easy	+
200% - 300°C	viscious (thick and sticky)	least volatile	only with a wick	large

TRENDS

- . The larger the molecules in a hydrocarbon
 - on the higher its boiling point
 - on the less volatile
 - on the less easy to flow / more viscous
 - on the less easy to burn

Retinery gos	1-4 carbon atoms	bottled gates, cooking, heating	25.
Giosoline (petrol)	5-6 carbon ofems	fuel for cors	1
Nachtna	6-10 carbon atoms	feedstack for chemicals and plastics	3-0707
Parattm (kerosene)	10-15 carbon along	fuel for directall, all stores, lamps	9-07
Diesel oil (gas oil)	15 -30 cerbon atoms	fuel for diesel engines	vTyce
Fuel oil 20-30 carbon of		A section of the sect	
Lubricating fraction	30-50 corbon atoms	oil for car engines, mathintry, water, polithes	29
Ritumen	> 50 carbon atoms	for road surfaces and roofs	

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K Morrison

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