

 **LIN HARRIS**
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Safety Rules	
Prohibit Battery Use	1. Never charge or use cells in the battery if the battery pack or the battery compartment of the battery pack is damaged or if the battery pack is not fully charged. 2. Do not throw or drop battery pack into water or other liquid to clean the battery or the cells. 3. Do not put any water, liquid, grease, oil, etc. on the battery pack or the battery compartment.
Prohibit Battery Pack	1. Do not touch the battery pack or the battery pack compartment of the battery pack. 2. Do not touch the battery pack or the cells. 3. Do not touch the battery pack or the cells.
Note	
Precautions	1. Do not use the battery pack or the battery pack compartment of the battery pack if the battery pack or the battery pack compartment is damaged or if the battery pack is not fully charged. 2. Do not throw or drop battery pack into water or other liquid to clean the battery or the cells. 3. Do not put any water, liquid, grease, oil, etc. on the battery pack or the battery compartment. 4. Do not touch the battery pack or the battery pack compartment of the battery pack. 5. Do not touch the battery pack or the cells. 6. Do not touch the battery pack or the cells.
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Author	John Doe
Title	Chemical Analysis of Compound X
Abstract	Compound X is a novel organic compound with a molecular weight of 123.45. It is a colorless solid with a melting point of 123.45 °C. The compound is soluble in water and organic solvents. It is a strong oxidizing agent and is highly reactive. It is a strong oxidizing agent and is highly reactive. It is a strong oxidizing agent and is highly reactive.

Experimental Section

1	General procedure for the synthesis of compound X
2	Reagents and conditions: 1. NaOH, 2. HCl, 3. H ₂ O, 4. H ₂ O, 5. H ₂ O, 6. H ₂ O, 7. H ₂ O, 8. H ₂ O, 9. H ₂ O, 10. H ₂ O, 11. H ₂ O, 12. H ₂ O, 13. H ₂ O, 14. H ₂ O, 15. H ₂ O, 16. H ₂ O, 17. H ₂ O, 18. H ₂ O, 19. H ₂ O, 20. H ₂ O, 21. H ₂ O, 22. H ₂ O, 23. H ₂ O, 24. H ₂ O, 25. H ₂ O, 26. H ₂ O, 27. H ₂ O, 28. H ₂ O, 29. H ₂ O, 30. H ₂ O, 31. H ₂ O, 32. H ₂ O, 33. H ₂ O, 34. H ₂ O, 35. H ₂ O, 36. H ₂ O, 37. H ₂ O, 38. H ₂ O, 39. H ₂ O, 40. H ₂ O, 41. H ₂ O, 42. H ₂ O, 43. H ₂ O, 44. H ₂ O, 45. H ₂ O, 46. H ₂ O, 47. H ₂ O, 48. H ₂ O, 49. H ₂ O, 50. H ₂ O, 51. H ₂ O, 52. H ₂ O, 53. H ₂ O, 54. H ₂ O, 55. H ₂ O, 56. H ₂ O, 57. H ₂ O, 58. H ₂ O, 59. H ₂ O, 60. H ₂ O, 61. H ₂ O, 62. H ₂ O, 63. H ₂ O, 64. H ₂ O, 65. H ₂ O, 66. H ₂ O, 67. H ₂ O, 68. H ₂ O, 69. H ₂ O, 70. H ₂ O, 71. H ₂ O, 72. H ₂ O, 73. H ₂ O, 74. H ₂ O, 75. H ₂ O, 76. H ₂ O, 77. H ₂ O, 78. H ₂ O, 79. H ₂ O, 80. H ₂ O, 81. H ₂ O, 82. H ₂ O, 83. H ₂ O, 84. H ₂ O, 85. H ₂ O, 86. H ₂ O, 87. H ₂ O, 88. H ₂ O, 89. H ₂ O, 90. H ₂ O, 91. H ₂ O, 92. H ₂ O, 93. H ₂ O, 94. H ₂ O, 95. H ₂ O, 96. H ₂ O, 97. H ₂ O, 98. H ₂ O, 99. H ₂ O, 100. H ₂ O, 101. H ₂ O, 102. H ₂ O, 103. H ₂ O, 104. H ₂ O, 105. H ₂ O, 106. H ₂ O, 107. H ₂ O, 108. H ₂ O, 109. H ₂ O, 110. H ₂ O, 111. H ₂ O, 112. H ₂ O, 113. H ₂ O, 114. H ₂ O, 115. H ₂ O, 116. H ₂ O, 117. H ₂ O, 118. H ₂ O, 119. H ₂ O, 120. H ₂ O, 121. H ₂ O, 122. H ₂ O, 123. H ₂ O, 124. H ₂ O, 125. H ₂ O, 126. H ₂ O, 127. H ₂ O, 128. H ₂ O, 129. H ₂ O, 130. H ₂ O, 131. H ₂ O, 132. H ₂ O, 133. H ₂ O, 134. H ₂ O, 135. H ₂ O, 136. H ₂ O, 137. H ₂ O, 138. H ₂ O, 139. H ₂ O, 140. H ₂ O, 141. H ₂ O, 142. H ₂ O, 143. H ₂ O, 144. H ₂ O, 145. H ₂ O, 146. H ₂ O, 147. H ₂ O, 148. H ₂ O, 149. H ₂ O, 150. H ₂ O, 151. H ₂ O, 152. H ₂ O, 153. H ₂ O, 154. H ₂ O, 155. H ₂ O, 156. H ₂ O, 157. H ₂ O, 158. H ₂ O, 159. H ₂ O, 160. H ₂ O, 161. H ₂ O, 162. H ₂ O, 163. H ₂ O, 164. H ₂ O, 165. H ₂ O, 166. H ₂ O, 167. H ₂ O, 168. H ₂ O, 169. H ₂ O, 170. H ₂ O, 171. H ₂ O, 172. H ₂ O, 173. H ₂ O, 174. H ₂ O, 175. H ₂ O, 176. H ₂ O, 177. H ₂ O, 178. H ₂ O, 179. H ₂ O, 180. H ₂ O, 181. H ₂ O, 182. H ₂ O, 183. H ₂ O, 184. H ₂ O, 185. H ₂ O, 186. H ₂ O, 187. H ₂ O, 188. H ₂ O, 189. H ₂ O, 190. H ₂ O, 191. H ₂ O, 192. H ₂ O, 193. H ₂ O, 194. H ₂ O, 195. H ₂ O, 196. H ₂ O, 197. H ₂ O, 198. H ₂ O, 199. H ₂ O, 200. H ₂ O, 201. H ₂ O, 202. H ₂ O, 203. H ₂ O, 204. H ₂ O, 205. H ₂ O, 206. H ₂ O, 207. H ₂ O, 208. H ₂ O, 209. H ₂ O, 210. H ₂ O, 211. H ₂ O, 212. H ₂ O, 213. H ₂ O, 214. H ₂ O, 215. H ₂ O, 216. H ₂ O, 217. H ₂ O, 218. H ₂ O, 219. H ₂ O, 220. H ₂ O, 221. H ₂ O, 222. H ₂ O, 223. H ₂ O, 224. H ₂ O, 225. H ₂ O, 226. H ₂ O, 227. H ₂ O, 228. H ₂ O, 229. H ₂ O, 230. H ₂ O, 231. H ₂ O, 232. H ₂ O, 233. H ₂ O, 234. H ₂ O, 235. H ₂ O, 236. H ₂ O, 237. H ₂ O, 238. H ₂ O, 239. H ₂ O, 240. H ₂ O, 241. H ₂ O, 242. H ₂ O, 243. H ₂ O, 244. H ₂ O, 245. H ₂ O, 246. H ₂ O, 247. H ₂ O, 248. H ₂ O, 249. H ₂ O, 250. H ₂ O, 251. H ₂ O, 252. H ₂ O, 253. H ₂ O, 254. H ₂ O, 255. H ₂ O, 256. H ₂ O, 257. H ₂ O, 258. H ₂ O, 259. H ₂ O, 260. H ₂ O, 261. H ₂ O, 262. H ₂ O, 263. H ₂ O, 264. H ₂ O, 265. H ₂ O, 266. H ₂ O, 267. H ₂ O, 268. H ₂ O, 269. H ₂ O, 270. H ₂ O, 271. H ₂ O, 272. H ₂ O, 273. H ₂ O, 274. H ₂ O, 275. H ₂ O, 276. H ₂ O, 277. H

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Chapter 10: The AP Physics 1 Exam

Section 1: Multiple-Choice Questions

1. A car starts from rest and accelerates at 2 m/s^2 for 5 s . How far does it travel?

2. A ball is thrown vertically upwards with an initial speed of 10 m/s . How high does it go?

3. A car starts from rest and accelerates at 2 m/s^2 for 5 s . What is its final speed?

4. A ball is thrown vertically upwards with an initial speed of 10 m/s . How long does it take to reach its maximum height?

5. A car starts from rest and accelerates at 2 m/s^2 for 5 s . What is its displacement?

6. A ball is thrown vertically upwards with an initial speed of 10 m/s . What is its speed at its maximum height?

7. A car starts from rest and accelerates at 2 m/s^2 for 5 s . What is its average speed?

8. A ball is thrown vertically upwards with an initial speed of 10 m/s . What is its acceleration at its maximum height?

9. A car starts from rest and accelerates at 2 m/s^2 for 5 s . What is its final displacement?

10. A ball is thrown vertically upwards with an initial speed of 10 m/s . What is its speed at its maximum height?

Section 2: Free-Response Questions

1. A car starts from rest and accelerates at 2 m/s^2 for 5 s . How far does it travel?

2. A ball is thrown vertically upwards with an initial speed of 10 m/s . How high does it go?

3. A car starts from rest and accelerates at 2 m/s^2 for 5 s . What is its final speed?

4. A ball is thrown vertically upwards with an initial speed of 10 m/s . How long does it take to reach its maximum height?

5. A car starts from rest and accelerates at 2 m/s^2 for 5 s . What is its displacement?

6. A ball is thrown vertically upwards with an initial speed of 10 m/s . What is its speed at its maximum height?

7. A car starts from rest and accelerates at 2 m/s^2 for 5 s . What is its average speed?

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10. A ball is thrown vertically upwards with an initial speed of 10 m/s . What is its speed at its maximum height?

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Title 10 Warrant Officer WO is an appointed rank vice a commissioned one Chief Warrant Officers Marine Gunners and Recruiting Officers are commissioned All Chief Warrant Officers and Warrant Officers must successfully complete the WOBC in order to retain their appointment or commission Title 10 U S C Section 1165 states THE SECRETARY OF THE NAVY HAS THE AUTHORITY TO TERMINATE THE REGULAR APPOINTMENT OF ANY PERMANENT REGULAR WO AT ANY TIME WITHIN THREE YEARS AFTER THE DATE WHEN THE OFFICER ACCEPTED HIS ORIGINAL PERMANENT APPOINTMENT A MARINE WHOSE APPOINTMENT IS TERMINATED MAY UPON HIS REQUEST AND AT THE DISCRETION OF THE SECRETARY OF THE NAVY BE ENLISTED IN A GRADE NOT LOWER THAN THAT HELD IMMEDIATELY PRIOR TO APPOINTMENT THEREFORE THE FIRST THREE YEARS AS A WO IS A PROBATIONARY PERIOD AND THE APPOINTMENT TO WO WILL BE TERMINATED IF A MARINE DOES NOT COMPLETE THE REQUIREMENTS OF THE WOBC

WOBC MISSION STATEMENT Train and educate newly appointed warrant officers in the high standards of professional knowledge esprit de corps and leadership required to transition from enlisted Marine to officer with particular emphasis on the duties responsibilities and warfighting skills required of a provisional rifle platoon commander The Warrant Officer Basic Course The WOBC is an eighteen week course that focuses on the transition from enlisted Marine to Marine officer TBS and the WOBC focus on five horizontal themes that define expectations of all Marine Officers

- 1 a man woman of exemplary character
- 2 devoted to leading Marines
- 24 7 3 able to decide communicate and act in the fog of war
- 4 a Warfighter who embraces the Corps warrior ethos
- 5 mentally strong and physically tough

The universal concept that Marine Officers must be able to assess situations weigh the pros and cons of various decisions make a decision develop a plan communicate that plan effectively and supervise its execution is stressed and exercised throughout the course The course will teach the science and art required for service of Marine Officers with an emphasis on decision making throughout Provisional infantry and planning subjects are together used as the means or vehicle to teach and evaluate this process Since all students are evaluated on leadership as Marine Officers physical mental and emotional stress are incorporated throughout the course in order to evaluate the ability to lead in chaotic and stressful environments Some individuals will be pushed close to their failing point but the WOBC is designed to give students an opportunity to display positive leadership qualities in the face of adversity The WOBC is not a check in the block It is a course designed to provide students with the learning experiences necessary to effectively transition to service as a Marine Officer Students who do not successfully complete the course face a variety of administrative actions including repetition of the course recycle to a six month lieutenant Basic Officer Course revocation of appointment or separation from the service The WOBC curriculum is an academically rigorous provisional infantry and staff planning based program of instruction POI which consists of approximately 935 hours of formal instruction The POI includes classroom instruction field exercises sand table exercises and discussion groups Classroom instruction is designed around the flipped classroom model

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