

CHAPTER 13: KINETICS: LEARNING GOALS

Learning Goal 1: Reaction Rates

Pages 13.1-13.2

Reaction rate is the change in concentration of a reactant or product over time.

Interpret the Equation: This is a rate of change equation. The equation shows how much faster the reaction will proceed as the concentration of a reactant or product increases.

Key Idea:

- For a reaction, rate can be expressed as:
 - Change in concentration of a reactant over time (negative value)
 - Change in concentration of a product over time (positive value)
 - Change in concentration of a reactant over time (negative value)
 - Change in concentration of a product over time (positive value)
 - Change in concentration of a reactant over time (negative value)
 - Change in concentration of a product over time (positive value)

Learning Goal 2:

A. Product of two functions

B. Rate laws

Product of two functions is the result of multiplying the two functions together. The product of two functions is the result of multiplying the two functions together. The product of two functions is the result of multiplying the two functions together.

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