

CHAPTER 21 REVIEW

Nuclear Chemistry**SECTION 1****SHORT ANSWER** Answer the following questions in the space provided.

- _____ Based on the information about the three elementary particles in the text, which has the greatest mass?
 - the proton
 - the neutron
 - the electron
 - They all have the same mass.
- _____ The force that keeps nucleons together is
 - a strong nuclear force.
 - a weak nuclear force.
 - an electromagnetic force.
 - a gravitational force.
- _____ The stability of a nucleus is most affected by the
 - number of neutrons.
 - number of protons.
 - number of electrons.
 - ratio of neutrons to protons.
- _____ If an atom should form from its constituent particles,
 - matter is lost and energy is taken in.
 - matter is lost and energy is released.
 - matter is gained and energy is taken in.
 - matter is gained and energy is released.
- _____ For atoms of a given mass number, those with greater mass defects, have
 - smaller binding energies per nucleon.
 - greater binding energies per nucleon.
 - the same binding energies per nucleon as those with smaller mass defects.
 - variable binding energies per nucleon.
- Based on **Figure 1.1** of the text, which isotope of He, helium-3 or helium-4,
 - _____ a. has the smaller binding energy per nucleon?
 - _____ b. is more stable to nuclear changes?
- The number of neutrons in an atom of magnesium-25 is _____.
- Nuclides of the same element have the same _____.

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