11-5

Study Guide and Intervention

7SDAPL3

Measures of Variation

The lower quartile or LQ is the median of the lower half of a set of data. The upper quartile or UQ is the median of the upper half of a set of data. The interquartile range is the difference between the upper quartile and the lower quartile.

Example 1

Find the range, median, upper and lower quartiles, and interquartile range for the following set of data. 13, 20, 18, 12, 21, 2, 18, 17, 15, 10, 14

The greatest number in the data set is 21. The least number is 2. The range is 21 - 2 or 19.

To find the quartiles, arrange the numbers in order from least to greatest.

2 10 12 13 14 15 17 18 18 20 21

The median is 15. The numbers below 15 are 2, 10, 12, 13, and 14. The median of the numbers below 15 is 12, so the lower quartile is 12. The numbers above 15 are 17, 18, 18, 20, and 21. The median of the numbers above 15 is 18, so the upper quartile is 18. The interquartile range is 18 - 12 or 6.

In some data sets, a few of the values are much greater than or less than the rest of the data. Data that are more than 1.5 times the value of the interquartile range beyond the quartiles are called outliers.

Example 2

Find any outliers for the set of data given in Example 1.

The interquartile range is 18-12 or 6. Multiply the interquartile range by 1.5.

 $6 \times 1.5 = 9$

Any data more than 9 above the upper quartile or below the lower quartile are outliers. Find the limits of the outliers.

Subtract 9 from the lower quartile. Add 9 to the upper quartile.

12 - 9 = 3 18 + 9 = 27

The limits of the outliers are 3 and 27. The only data point outside this range is 2, so the only outlier is 2.

Exercises

Find the range, median, upper and lower quartiles, interquartile range, and any outliers for each set of data.

1. 14, 16, 18, 24, 19, 15, 13

2, 29, 27, 24, 28, 30, 51, 28

3. 57, 60, 43, 55, 46, 43, 62, 31

4. 91, 92, 88, 89, 93, 95, 65, 85, 91

5. 104, 116, 111, 108, 113, 127, 109, 122, 115, 105

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Susan Pauline O'Hara, Maureen McMahon

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Teaching Middle School Mathematics Douglas K. Brumbaugh, 2013-05-13 Middle school teaching and learning has a distinct pedagogy and curriculum that is grounded in the concept of developmentally appropriate education This text is designed to meet the very specific professional development needs of future teachers of mathematics in middle school environments Closely aligned with the NCTM Principles and Standards for School Mathematics the reader friendly interactive format encourages readers to begin developing their own teaching style and making informed decisions about how to approach their future teaching career A variety of examples establish a broad base of ideas intended to stimulate the formative development of concepts and models that can be employed in the classroom Readers are encouraged and motivated to become teaching professionals who are lifelong learners. The text offers a wealth of technology related information and activities reflective thought provoking questions mathematical challenges student life based applications TAG tricks activities games sections and group discussion prompts to stimulate each future teacher's thinking Your Turn sections ask readers to work with middle school students directly in field experience settings This core text for middle school mathematics methods courses is also appropriate for elementary and secondary mathematics methods courses that address teaching in the middle school grades and as an excellent in service resource for aspiring or practicing teachers of middle school mathematics as they update their knowledge base Topics covered in Teaching Middle School Mathematics NCTM Principles for School Mathematics Representation Connections Communication Reasoning and Proof Problem Solving Number and Operations Measurement Data Analysis and Probability Algebra in the Middle School Classroom and Geometry in the Middle School Classroom New York Review Series: Grade 8 Mathematics Review Workbook McGraw-Hill Education, 2005-11-21 New York Review Series Grade 8 Mathematics Review helps students succeed on the New York 8th grade test Students review both Post March and Pre March topics Lessons for each performance indicator include fully worked out examples and exercises that are similar to those on the test Additional problem solving lessons as well as chapter tests and practice tests are included **Entering the World of Work, Student Edition** McGraw-Hill, 2006-01-01 Created to help students with special needs develop the skills they need to succeed in their careers and in life Entering the World of Work focuses on career issues money management and balancing work and personal life Daily Math Thinking Routines in Action Nicki Newton, 2018-09-05 Bring math to life with routines that are academically rigorous standards based and engaging Go beyond circling ABCD on your bell ringers and do nows and get your students reasoning modeling and communicating about math every day In this new book from bestselling author and consultant Dr Nicki Newton you ll learn how to develop effective daily routines to improve students thinking reasoning and questioning about math The book provides a wide variety of rigorous high interest routines and explains how to rotate and implement them into your curriculum Inside you ll find Questioning techniques that encourage students to think beyond the right vs wrong continuum

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the objectives of the boiling point determination lab were to determine the boiling point of a mystery substance provided to each student boiling point is the point at which a liquid changes physical states and becomes a vapor or gas boiling point is important to understand because it can be used to help distinguish between different substances

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the boiling point of an organic liquid is the temperature at which its vapour pressure equals the atmospheric pressure over the liquid or it is the temperature at which the vapour and liquid phases are in equilibrium at agiven pressure the boiling point is considered as a criterion of purity of a compound and is useful for identification of

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given that the boiling points of ethanol and methanol 78 c and 64 c respectively are far off in a significant way this seems like a valid conclusion conclusion thermometer calibrations are necessary in order to collect reliable information as precise calculations are imperative to gathering correct results

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melting point 1 a push the open end of a capillary tube into the powdered acetamide b move the powder to the closed end of the capillary tube by tapping it on the table repeat until the the powdered acetamide occupies 1 2 mm of the capillary tube end c

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discussion and conclusion the technique recrystallization was used because it is a simple method to purify a solid compound the percentage yield for benzoic acid and acetanilide acid being high indicated the initial sample had a low concentration of impurities with the melting point being in a specific

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introduction in this lab we will be using boiling points melting points and infrared spectrometry to identify unknown compounds boiling points are affect by three key factors the first being intermolecular forces inter molecular affect boiling point based on bond strength

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determination of boiling points introduction the boiling point of a compound is the temperature at which it changes from a liquid to a gas this is a physical property often used to identify substances or to check the purity of the compound it **boiling point determination lab report introduction the** - Jul 19 2023

clean up solid and liquid spills immediately true g conclusions after conducting the boiling point determination lab it can be concluded that the unknown substance 193 is isopropyl alcohol on account of its observed boiling point of 88 c

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conclusion after completing this experiment the melting point of three different unknown compounds was obtained using the mel temp apparatus compound a had a melting point range of 114 6 115 5 c thus revealing it to be acetanilide compound b had a melting point range of 122 6 124 5 c thus

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nov 27 2019 essay sample the purpose of this experiment is to determine the boiling points of various organic compounds and to use these to identify unknowns equipment free essays

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melting point can be used as the identity and purity of an organic solid however it is not the same as the boiling point the

boiling point is a characteristic unique of pure liquids the boiling point is the temperature at which the vapor pressure of a liquid is equal to the atmospheric pressure

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aug 11 2020 boiling points are very sensitive to changes in applied pressure so all boiling points should be reported with the measured pressure a compound s normal boiling point refers to its boiling point at a pressure of 760 text mm ce hg **experiment 11 vapour pressure proctech 2ce3 lab** - Jul 07 2022

the purpose of the experiment is to understand what vapour pressure is and how it changes with temperature to understand the relationship between vapour pressure at a given temperature and boiling temperature at a given pressure to determine the vapour pressure of a pure liquid at various temperatures

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exploring boiling points abstract in this experiment we are given a set of straight chain alkanes and a set of compounds tocompare by following the trendline of the straight chain alkanes it could be concluded that as the molar mass of the straight chain alkanes increases their boiling points increase as well

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laboratory partners samantha and mackenzie goals \square correctly determine the boiling point of an unknown liquid \square measure the refractive index of an unknown \square use the gas chromatogram to determine the mass percentage of a mixture and boiling point determination faculty of science - Jun 18 2023

50 c so in order to convert an experimental measurement taken in calgary which is at higher altitude than sea level and so is at a lower pressure to that reported for sea level higher pressure one needs to add a correction factor since the boiling point at sea level is higher than that at higher altitudes

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introduction in chemistry one may encounter an unknown substance that need to be determined by physical properties such as boiling point melting point density solubility etc in this experiment i am to figure out the boiling point of a mystery substance in order to identify it chemical responsibility this experiment requires the handling of a bunsen burner boiling 6 2b step by step procedures for boiling point determination - Mar 15 2023

there are a variety of methods by which a sample s boiling point can be determined including distillation reflux and by using

a thiele tube the most straightforward method uses a thiele tube and has the advantage of using less than 0 5ml 0 5 ml of
material
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