Nacros:	Photograph 1	
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Student Exploration: Stoichiometry

Giarreso Winares san-

First as a coals follows a recipe to decide how much of each ingredient to add, a chemist uses **stolchowntry** to determine the amounts of substances involved in chemical reactions. The Stolchowntry Gramo^{na} allows you to try much hand at figuring out the amounts of macrans and products that take part in a chemical reaction.

To begin, check that this equation is shown.

Look at the **coefficients** (such as the "3" in SCC) is from of each substance in the equation. The
coefficients tell you have many reducutes or atoms take part in a chemical reaction. In the spaces below,
list the number of each reclicule or atom in the equation:

Fe₂O₃ ____ CO_{___} Fe_{___} CO₁___

In a balanced equation. The same number of each kind of atom is shown on each side of the equation.
 Calculate the number of each 0°co, oxygen (C), and carbon atoms (C).

 Reaction is
 Incom.
 Couper
 Carbon.

 Products
 Ison.
 Dogger
 Carbon.

based on these values, is the equation balanced?

Activity As

Microson.

Get the Girns ready Check that the equation is still. FeuD₁ + 3CO → 2Fe + 3CO₂

Hirson, slick New equation until it recopposes.



Question: How do scientists find the molecular mass and replat mass of a substance?

Calculate: The **restrictive mass** of a recreate is the sum of the masses of each atom in the molecule. The unit of molecule is grants.

from's attentic tracks is 55.5g, carbon's mass is 12.0g, and paygen's mass is 16.0g.

Calculate the molecular mass of carbon monoxide (CC) by adding the atomic mass of carbon and the atomic mass of oxygen:

Calculate the molecular mass of carbon dioxide (CO₄): _

Calculate the molecular mass of fron HID code (Fe/Cyt

Check your answers on the Cizmo by inspecting the middle row of tiles on the right side of the Cizmo. These tiles show the units "1, mol" on top and "g" below.

Gizmo Warm Up Answers

Katrin Zwirglmaier



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