

Name: Ahmad Hamoud Date: 2/11/2020

## Student Exploration: Electron Configuration

**Vocabulary:** atomic number, atomic radius, Aufbau principle, chemical family, diagonal rule, electron configuration, Hund's rule, orbital, Pauli exclusion principle, period, shell, spin, sub-shell

**Prior Knowledge Questions** (Do these BEFORE using the Gizmo.)

1. Elvis Perkins, a rather shy fellow, is getting on the bus shown at right. Which seat do you think he will probably sit in? Mark this seat with an "E."
2. Marta Warren gets on the bus after Elvis. She is tired after a long day at work. Where do you think she will sit? Mark this seat with an "M."
3. In your experience, do strangers getting on a bus like to sit with other people if there is an empty seat available? NO



### Gizmo Warm-up

Just like passengers getting on a bus, electrons orbit the nuclei of atoms in particular patterns. You will discover these patterns (and how electrons sometimes act like passengers boarding a bus) with the *Electron Configuration Gizmo*.

To begin, check that **Lithium** is selected on the PERIODIC TABLE tab.

1. The **atomic number** is equal to the number of protons in an atom.

How many protons are in a lithium atom? 3

2. A neutral atom has the same number of electrons and protons.

How many electrons are in a neutral lithium atom? 3

3. Select the ELECTRON CONFIGURATION tab, and check that **Energy** is selected. Click twice in the **1s** box at lower left and once in the **2s** box. Observe the atom model at right.

A. What do you see? a nucleus with somethings orbiting it



3	4	5	6	7	8
Sc	Ti	V	Cr	Mn	Fe
Y	Zr	Nb	Mo	Tc	Ru
La	Hf	Ta	W	Re	Os



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## **Table of Contents Gizmo Student Exploration Electron Configuration Answers Key**

1. Understanding the eBook Gizmo Student Exploration Electron Configuration Answers Key
  - The Rise of Digital Reading Gizmo Student Exploration Electron Configuration Answers Key
  - Advantages of eBooks Over Traditional Books
2. Identifying Gizmo Student Exploration Electron Configuration Answers Key
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Gizmo Student Exploration Electron Configuration Answers Key
  - User-Friendly Interface
4. Exploring eBook Recommendations from Gizmo Student Exploration Electron Configuration Answers Key
  - Personalized Recommendations
  - Gizmo Student Exploration Electron Configuration Answers Key User Reviews and Ratings
  - Gizmo Student Exploration Electron Configuration Answers Key and Bestseller Lists
5. Accessing Gizmo Student Exploration Electron Configuration Answers Key Free and Paid eBooks

- Gizmo Student Exploration Electron Configuration Answers Key Public Domain eBooks
  - Gizmo Student Exploration Electron Configuration Answers Key eBook Subscription Services
  - Gizmo Student Exploration Electron Configuration Answers Key Budget-Friendly Options
6. Navigating Gizmo Student Exploration Electron Configuration Answers Key eBook Formats
- ePub, PDF, MOBI, and More
  - Gizmo Student Exploration Electron Configuration Answers Key Compatibility with Devices
  - Gizmo Student Exploration Electron Configuration Answers Key Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Gizmo Student Exploration Electron Configuration Answers Key
  - Highlighting and Note-Taking Gizmo Student Exploration Electron Configuration Answers Key
  - Interactive Elements Gizmo Student Exploration Electron Configuration Answers Key
8. Staying Engaged with Gizmo Student Exploration Electron Configuration Answers Key
- Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Gizmo Student Exploration Electron Configuration Answers Key
9. Balancing eBooks and Physical Books Gizmo Student Exploration Electron Configuration Answers Key
- Benefits of a Digital Library
  - Creating a Diverse Reading Collection Gizmo Student Exploration Electron Configuration Answers Key
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Gizmo Student Exploration Electron Configuration Answers Key
- Setting Reading Goals Gizmo Student Exploration Electron Configuration Answers Key
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Gizmo Student Exploration Electron Configuration Answers Key
- Fact-Checking eBook Content of Gizmo Student Exploration Electron Configuration Answers Key
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development

- Exploring Educational eBooks

### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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