



Name: Aaeysa M

Date:

Student Exploration: Titration

Directions: Follow the instructions to go through the simulation. Respond to the questions and prompts in the orange boxes.

Vocabulary: acid, analyte, base, dissociate, equivalence point, indicator, litmus paper, molarity, neutralize, pH, strong acid, strong base, titrant, titration, titration curve, weak acid, weak base

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

There are several definitions of acids and bases. According to the Brønsted-Lowry definition, an **acid** is a substance that is capable of donating a proton to another substance. A **base** is a substance that accepts protons. When an acid and a base are combined, the acid is **neutralized** as the base accepts the protons produced by the acid.



One way to determine if a solution is acidic or basic is to use **litmus paper**, as shown above. There are two types of litmus papers: red and blue.

How does litmus paper indicate an acid?

Both strips turn red.

How does litmus paper indicate a neutral substance?

The red strip stays red and the blue one stays blue.

How does litmus paper indicate a base?

Both strips turn blue.

Gizmo Warm-up

Litmus is an example of an **indicator**, a substance that changes color depending on its **pH** (pH is a measure of the concentration of protons, or H^+ ions). In the Titration Gizmo, you will use indicators to show how acids are neutralized by bases, and vice versa.

To begin, check that **1.00 M NaOH** is selected for the **Burette**, **Mystery HBr** is selected for the **Flask**, and **Bromthymol blue** is selected for the **Indicator**.



1. Look at the flask. What is the color of the bromthymol blue indicator?

yellow

2. What does this tell you about the pH of the solution in the flask?

pH is below 6.0

Solutions with a pH below 7.0 are acidic, while those with a pH above 7.0 are basic.

3. Move the slider on the burette to the top to add about 25 mL of NaOH to the flask. What happens, and what does this tell you about the pH of the flask?

The solution turns blue meaning the pH is now above 7.0.

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Yijin Wang



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Table of Contents Gizmo Answer Keys Explore Learning

1. Understanding the eBook Gizmo Answer Keys Explore Learning
 - The Rise of Digital Reading Gizmo Answer Keys Explore Learning
 - Advantages of eBooks Over Traditional Books
2. Identifying Gizmo Answer Keys Explore Learning
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Gizmo Answer Keys Explore Learning
 - User-Friendly Interface

4. Exploring eBook Recommendations from Gizmo Answer Keys Explore Learning
 - Personalized Recommendations
 - Gizmo Answer Keys Explore Learning User Reviews and Ratings
 - Gizmo Answer Keys Explore Learning and Bestseller Lists
5. Accessing Gizmo Answer Keys Explore Learning Free and Paid eBooks
 - Gizmo Answer Keys Explore Learning Public Domain eBooks
 - Gizmo Answer Keys Explore Learning eBook Subscription Services
 - Gizmo Answer Keys Explore Learning Budget-Friendly Options
6. Navigating Gizmo Answer Keys Explore Learning eBook Formats
 - ePub, PDF, MOBI, and More
 - Gizmo Answer Keys Explore Learning Compatibility with Devices
 - Gizmo Answer Keys Explore Learning Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Gizmo Answer Keys Explore Learning
 - Highlighting and Note-Taking Gizmo Answer Keys Explore Learning
 - Interactive Elements Gizmo Answer Keys Explore Learning
8. Staying Engaged with Gizmo Answer Keys Explore Learning
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Gizmo Answer Keys Explore Learning
9. Balancing eBooks and Physical Books Gizmo Answer Keys Explore Learning
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Gizmo Answer Keys Explore Learning
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Gizmo Answer Keys Explore Learning
 - Setting Reading Goals Gizmo Answer Keys Explore Learning
 - Carving Out Dedicated Reading Time

12. Sourcing Reliable Information of Gizmo Answer Keys Explore Learning
 - Fact-Checking eBook Content of Gizmo Answer Keys Explore Learning
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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