

## L'Etranger d'Albert Camus Questionnaire de lecture

1. Le dispositif énonciatif mis en place dans l'incipit conduit le lecteur à ☐ considérer Meursault comme un étranger ☐ se mettre dans la peau d'un étranger ☐ considérer le monde comme étranger
2. Au total le roman comporte ☐ 8 chapitres ☐ 9 chapitres ☐ 11 chapitres ☐ 1 chapitre
3. Camus a songé à plusieurs titres pour son roman. Cochez l'intrus ? ☐ « Un Homme pressé » ☐ « La Pudeur » ☐ « Un Homme heureux » ☐ « Un Homme libre » ☐ « Un Homme comme les autres »
4. L'adjectif « absurde » apparaît une seule fois dans le roman. Dans quel chapitre ? 1,3 ☐ 1,5 ☐ 1,6 ☐ 11,1 ☐ 11,3 ☐ 11,5
5. Meursault est informé du décès de sa mère ☐ par un sms ☐ par un appel téléphonique ☐ par un télégramme ☐ par une lettre
6. Que dit en substance ce message ? ☐ « Mère décédée. Enterrement demain. Sentiments distingués » ☐ « Mère décédée. Enterrement demain. Sincères condoléances » ☐ « Mère décédée ce jour. Enterrement demain. Sentiments distingués » ☐ « Mère décédée. Inhumation demain à maison de retraite. Sentiments distingués »
7. Les indications spatio-temporelles confèrent à cet incipit des allures de ☐ roman policier ☐ conte ☐ journal
8. Pourquoi le récit de l'annonce de la mort de la mère et des heures qui suivent est-il déconcertant ? ☐ Meursault ne manifeste aucun chagrin ☐ Meursault se rend dans une église ☐ Meursault est terrassé par le chagrin ☐ Meursault ne change rien à ses habitudes (2 réponses attendues)
9. Parlant de la mort de sa mère, Meursault utilise des termes tels que « jours de congé », « excuse », « affaire classée ». Que traduisent ces termes ? ☐ un esprit de fonctionnaire ☐ un amour filial ☐ de l'indifférence ☐ de la désinvolture
10. Ces termes semblent faire écho ☐ au sec langage administratif ☐ à la langue épurée du télégramme ☐ au code civil (2 réponses attendues)
11. Par quels termes ou gestes Meursault traduit-il son émotion ? ☐ « Maman » ☐ aucun terme du vocabulaire psychologique, susceptible de traduire une émotion quelconque, n'est mobilisé ☐ il déchire rageusement le télégramme ☐ il s'isole au balcon et pleure ☐ il fume une cigarette
12. Face aux formules stéréotypées du télégramme, Meursault laisse tomber « cela ne veut rien dire ». Comment comprendre ce jugement ? ☐ c'est absurde, ce style télégraphique occulte la destinée tragique des hommes dont l'horizon est la mort ☐ que peut bien vouloir signifier « demain », dès lors que le télégramme n'est pas daté ? « C'était peut-être hier » ! ☐ on ne peut résumer une vie en quelques mots ☐ Meursault ne comprend pas le langage administratif (2 réponses attendues)
13. Dans sa conversation avec son patron, Meursault a le sentiment que ce dernier éprouve de la mauvaise humeur face à sa demande et il perçoit dans le silence de son patron une forme de désapprobation. Dans ce contexte, que suggère la phrase « je lui ai même dit : ce n'est pas de ma faute » ? ☐ un effort de justification qui vise à le disqualifier d'une accusation implicite. ☐ le désir de s'attirer la sympathie et la compréhension du patron ☐ l'expression d'une forte culpabilité
14. Quelle pensée Meursault semble prêter à son patron ? ☐ il cherche une excuse pour avoir deux jours de congé ☐ il n'a pas du tout l'attitude d'un fils meurtri par la mort de sa mère ☐ c'est un comédien
15. Le seul sentiment, qui, finalement, transparait dans cet incipit, c'est ☐ le sentiment du devoir accompli ☐ le sentiment de culpabilité ☐ le sentiment filial
16. C'est à travers ce sentiment que se ressent le mieux l'affection de Meursault pour sa mère ☐ l'avoir mise à l'asile, c'est pour lui la certitude qu'elle était plus heureuse et il a peur d'en douter ☐ il regrette de l'avoir mise à l'asile ☐ il regrette de ne pas lui avoir rendu visite plus souvent
17. L'incipit peut être lu comme l'image d'un homme qui cherche à se persuader, lui et les autres, qu'il fait en sorte que ☐ sa mère le laisse vivre sa vie ☐ sa mère ne devienne pas une charge pour lui ☐ sa mère meure heureuse.
18. Tous les éléments du procès qui seront retenus contre Meursault se trouvent, finalement ☐ concentrés dans cet incipit ☐ absents de cet incipit

# Leacutetranger Dalbert Camus Questionnaire De Lecture

**Raffaela Di Napoli**



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$\text{HNO}_3 + \text{Mg}(\text{C}_2\text{H}_3\text{O}_2)_2 \rightarrow \text{H}_2\text{C}_2\text{H}_3\text{O}_2 + \text{Mg}(\text{NO}_3)_2$  chemical equation - Aug 11 2023

web definitions practice chemistry quizzes help chat forums enenglish login  $\text{HNO}_3 + \text{Mg}(\text{C}_2\text{H}_3\text{O}_2)_2 \rightarrow \text{H}_2\text{C}_2\text{H}_3\text{O}_2 + \text{Mg}(\text{NO}_3)_2$  chemical equation balancer balanced chemical equation  $2\text{HNO}_3 + \text{Mg}(\text{C}_2\text{H}_3\text{O}_2)_2 \rightarrow 2\text{H}_2\text{C}_2\text{H}_3\text{O}_2 + \text{Mg}(\text{NO}_3)_2$

**how to balance  $\text{Mg}(\text{OH})_2 + \text{HNO}_3 \rightarrow \text{Mg}(\text{NO}_3)_2 + \text{H}_2\text{O}$  magnesium youtube** - Nov 02 2022

web jun 10 2018 527 75k views 5 years ago in this video we'll balance the equation  $\text{Mg}(\text{OH})_2 + \text{HNO}_3 \rightarrow \text{Mg}(\text{NO}_3)_2 + \text{H}_2\text{O}$  and provide the correct coefficients for each compound more more

$\text{Mg} + \text{HNO}_3 \rightarrow \text{Mg}(\text{NO}_3)_2 + \text{H}_2$  chemical equation balancer - Oct 13 2023

web magnesium nitric acid magnesium nitrate dihydrogen  $\text{Mg} + \text{HNO}_3 \rightarrow \text{Mg}(\text{NO}_3)_2 + \text{H}_2$  is a single displacement substitution reaction where one mole of solid magnesium mg and two moles of aqueous nitric acid  $\text{HNO}_3$  react to form one mole of

aqueous magnesium nitrate  $\text{Mg(NO}_3)_2$  and one mole of dihydrogen  $\text{H}_2$  gas show chemical

**how to balance  $\text{Mg(NO}_3)_2 + \text{H}_2$**  breslyn.org - Jun 09 2023

web when balancing chemical equations our goal is to have the same number of each type of atom on both sides of the equation only change the coefficients these are the numbers in front substances never change the subscripts the small numbers after elements

how do you balance  $\text{Mg(NO}_3)_2 + \text{H}_2$  socratic - Mar 06 2023

web mar 4 2018  $\text{Mg(NO}_3)_2 + \text{H}_2$  on the left side there are Mg 1 magnesium in elemental form  $\text{H}_2$  there are 2 molecules of nitric acid each acid has one hydrogen so two molecules of nitric acid is 2 hydrogens N 2 there are 2 molecules of nitric acid each acid has one nitrogen so two molecules of nitric acid is 2 nitrogens O 6

solved 1 enter balanced molecular equation for the reaction - Apr 26 2022

web chemistry questions and answers 1 enter balanced molecular equation for the reaction between nitric acid and magnesium hydroxide express your answer as a chemical equation including phases 2 enter a net ionic equation for the reaction between nitric acid and magnesium hydroxide express your answer as a chemical equation including

$\text{MgCO}_3 + \text{HNO}_3 \rightarrow \text{Mg(NO}_3)_2 + \text{H}_2\text{O} + \text{CO}_2$  chemical equation - Jul 10 2023

web magnesium carbonate nitric acid magnesium nitrate water carbon dioxide one mole of magnesium carbonate  $\text{MgCO}_3$  and two moles of nitric acid  $\text{HNO}_3$  react to form one mole of magnesium nitrate  $\text{Mg(NO}_3)_2$  one mole of water  $\text{H}_2\text{O}$  and one mole of carbon dioxide  $\text{CO}_2$  show chemical structure image reactants

**how do you balance  $\text{Mg(NO}_3)_2 + \text{H}_2$**  byjus - Feb 05 2023

web magnesium is already balanced on the reactant and product side balancing nitrogen by multiplying  $\text{HNO}_3$  by 2 we get Mg s magnesium 2  $\text{HNO}_3$  l nitric acid  $\text{Mg(NO}_3)_2$  s magnesium nitrate  $\text{H}_2$  g hydrogen gas oxygen and hydrogen get automatically balanced by balancing nitrogen

*$\text{Mg(NO}_3)_2 + \text{H}_2\text{SO}_4 \rightarrow \text{MgSO}_4 + \text{HNO}_3$  chemical equation* - Apr 07 2023

web magnesium nitrate sulfuric acid magnesium sulfate nitric acid

*give the balanced chemical equation for the following reaction* - Aug 31 2022

web the balanced net ionic equation for the reaction of magnesium carbonate with nitric acid is a  $\text{MgCO}_3$  s 2h aq  $\text{Mg}^{2+}$  aq  $\text{CO}_2$  g  $\text{H}_2\text{O}$  l b  $\text{Mg}^{2+}$  aq  $\text{CO}_3^{2-}$  aq 2h aq 2 $\text{NO}_3^-$  aq Mg write the balanced equation when nitric acid reacts with calcium hydroxide to produce water and calcium nitrate

**magnesium with nitric acid equation** thefitnessmanual - Feb 22 2022

web magnesium oxide nitric acid gives rise to a magnesium phosphate hydrogen balanced equation on may 27 2019 how do you balance magnesium and nitric acid you ll need to count all of atoms on both directions of the chemical equation to

balance mg hno<sub>3</sub> mm no<sub>3</sub> 2 no<sub>2</sub> h<sub>2</sub>o

**magnesium and nitric acid balanced equation** - Oct 01 2022

web magnesium and nitric acid balanced equation general chemistry oct 29 2021 the eleventh edition was carefully reviewed with an eye toward strengthening the content available in owl2 end of chapter questions and updating the presentation nomenclature changes and the

*how to balance hno<sub>3</sub> mg mg no<sub>3</sub> 2 h<sub>2</sub> breslyn org* - Sep 12 2023

web how to balance hno<sub>3</sub> mg mg no<sub>3</sub> 2 h<sub>2</sub> word equation nitric acid magnesium magnesium nitrate hydrogen gas type of chemical reaction for this reaction we have a single displacement reaction balancing strategies in this single displacement reaction the mg replaces the h in the hno<sub>3</sub>

*how to balance mg hno<sub>3</sub> mg no<sub>3</sub> 2 no<sub>2</sub> h<sub>2</sub>o magnesium youtube* - May 08 2023

web nov 20 2020 in this video we ll balance the equation mg hno<sub>3</sub> mg no<sub>3</sub> 2 no<sub>2</sub> h<sub>2</sub>o and provide the correct coefficients for each compound to balance mg hno<sub>3</sub> mg no<sub>3</sub> 2 no<sub>2</sub> h<sub>2</sub>o you ll need to be

**lecture 13 acid base reactions widener university** - Dec 03 2022

web magnesium hydroxide nitric acid balanced total equation mg oh 2 aq 2 hno<sub>3</sub> aq 2 h<sub>2</sub> o mg no<sub>3</sub> 2 aq total ionic equation mg 2 2 oh 1 2 h 1 2 no<sub>3</sub> 1 2 h<sub>2</sub> o mg 2 2 no<sub>3</sub> 1 net ionic equation 2 oh 1 2 h 1 2 h<sub>2</sub> o sulfuric acid and lithium hydroxide balanced total equation h<sub>2</sub> so<sub>4</sub> aq 2 lihoh aq 2 h<sub>2</sub>