Name:	Date:	
	La value	

#### Student Exploration: Ionic Bonds

Directions: Follow the instructions to go through the simulation. Respond to the questions and prompts in the orange boxes.

Vocabulary: chemical family, electron affinity, ion, ionic bond, metal, nonmetal, octet rule, shell, valence electron

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

 Nate and Clara are drawing pictures with markers. There are 8 markers in a set. Nate has 9 markers and Clara has 7. What can Nate and Clara do so that each of them has a full set?

Nate can give Clara a marker.

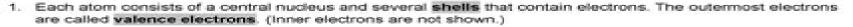
Maggie is sitting at a table with Fred and Florence. Maggie has 10 markers, but Fred and Florence each have only 7 markers. How can they share markers so each has 8?

Maggie can give Fred and Florence each a marker.

#### Gizmo Warm-up

Just like students sharing markers, atoms sometimes share or swap electrons. By doing this, atoms form bonds. The *lonic Bonds* Gizmo allows you to explore how **ionic bonds** form.

To begin, check that **Sodium (Na)** and **Chlorine (CI)** are selected from the menus at right. Click **Play** ( ) to see electrons orbiting the nucleus of each atom. (Note: These atom models are simplified and not meant to be realistic.)



How many valence electrons does each atom have? Sodium: 1 Chlorine: 7

- Click Pause ( !!!). Elements can be classified as metals and nonmetals. Metals do not hold on to their valence electrons very tightly, while nonmetals hold their electrons tightly. Electron affinity is a measure of how tightly the valence electrons are held.
  - A. Try pulling an electron away from each atom. Based on this experiment, which atom is a metal?

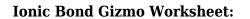
Sodium Which is a nonmetal? Chlorine

B. Try moving an electron from the metal to the nonmetal. What happens?

The nonmetal takes in the electron

# **Ionic Bond Gizmo Worksheet**

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#### **Table of Contents Ionic Bond Gizmo Worksheet**

- 1. Understanding the eBook Ionic Bond Gizmo Worksheet
  - The Rise of Digital Reading Ionic Bond Gizmo Worksheet
  - Advantages of eBooks Over Traditional Books
- 2. Identifying Ionic Bond Gizmo Worksheet
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Ionic Bond Gizmo Worksheet
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Ionic Bond Gizmo Worksheet
  - Personalized Recommendations
  - Ionic Bond Gizmo Worksheet User Reviews and Ratings

- Ionic Bond Gizmo Worksheet and Bestseller Lists
- 5. Accessing Ionic Bond Gizmo Worksheet Free and Paid eBooks
  - Ionic Bond Gizmo Worksheet Public Domain eBooks
  - Ionic Bond Gizmo Worksheet eBook Subscription Services
  - Ionic Bond Gizmo Worksheet Budget-Friendly Options
- 6. Navigating Ionic Bond Gizmo Worksheet eBook Formats
  - o ePub, PDF, MOBI, and More
  - Ionic Bond Gizmo Worksheet Compatibility with Devices
  - Ionic Bond Gizmo Worksheet Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - o Adjustable Fonts and Text Sizes of Ionic Bond Gizmo Worksheet
  - Highlighting and Note-Taking Ionic Bond Gizmo Worksheet
  - Interactive Elements Ionic Bond Gizmo Worksheet
- 8. Staying Engaged with Ionic Bond Gizmo Worksheet
  - o Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Ionic Bond Gizmo Worksheet
- 9. Balancing eBooks and Physical Books Ionic Bond Gizmo Worksheet
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Ionic Bond Gizmo Worksheet
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Ionic Bond Gizmo Worksheet
  - Setting Reading Goals Ionic Bond Gizmo Worksheet
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Ionic Bond Gizmo Worksheet
  - Fact-Checking eBook Content of Ionic Bond Gizmo Worksheet
  - Distinguishing Credible Sources

- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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